

Curso de Verão

Departamento de Física — UFPE

2024

Recife-PE

22 de Janeiro a **09 de Fevereiro**



Minicurso 3:

Materiais Bidimensionais

Prof. Lídia C. Gomes

Dia 1 (29/01)

General introduction.

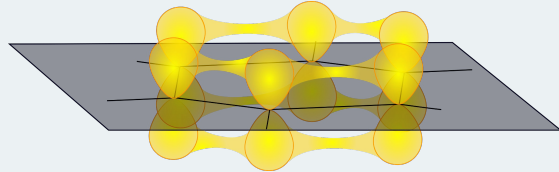
What are 2D materials?

A tiny bit of history.

Some interesting properties.

Dia 2 (30/01)

The origin of dimensionality: an analysis of orbital hybridization in graphene.



Dia 3 (31/01)

Synthesis methods.

Some experimental achievements.

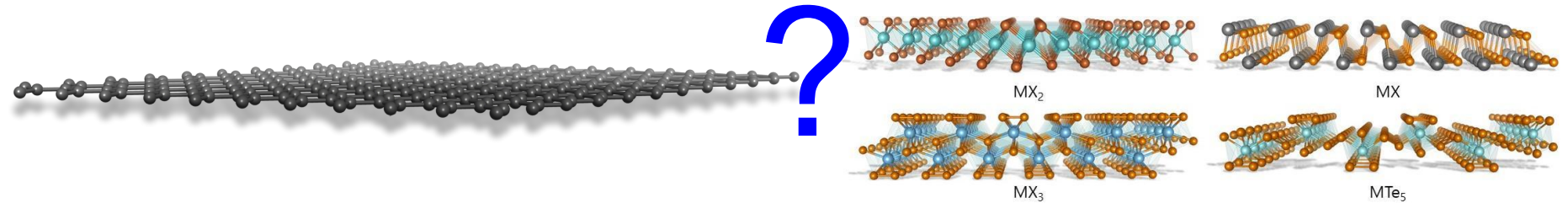
Some help from computers: ML discovery and development of new 2D materials.

Materiais bidimensionais

O que são?

Cristais com um ou poucos átomos de espessura

Interações eletrônicas no plano são muito mais fortes do que aquelas na direção perpendicular ao plano.



Approaches

Top-down

mechanical exfoliation

liquid exfoliation

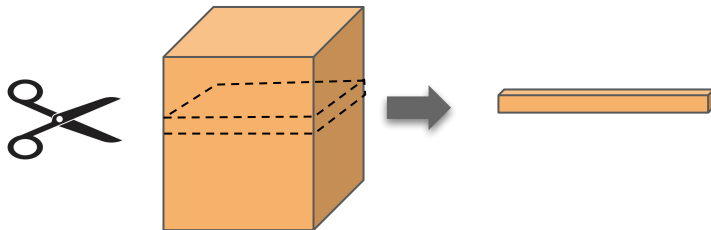
ultrasonic exfoliation

electrochemical exfoliation

ionchange exfoliation

lithium intercalation exfoliation

chemical reduction



Bottom-up

epitaxial growth

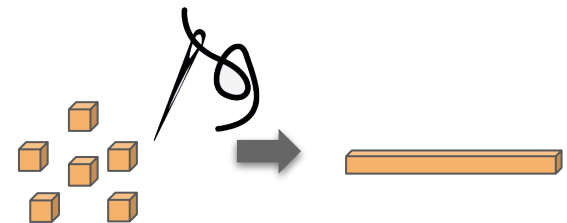
chemical vapor deposition (CVD)

pulsed laser deposition

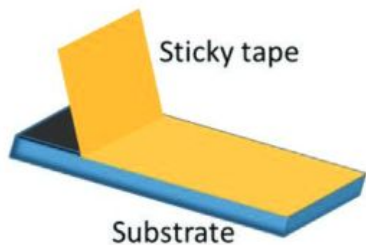
wet chemical methods

microwave assisted method

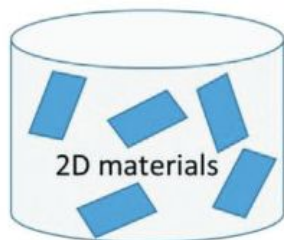
topochemical transformation



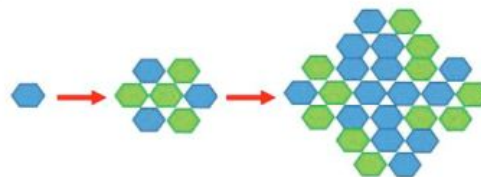
Mechanical cleavage



Liquid vapour exfoliation

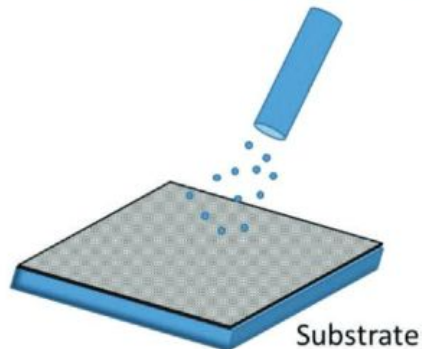


Chemical synthesis

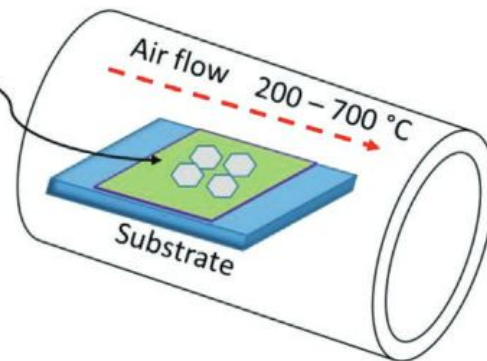
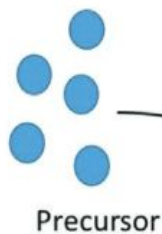


Metal substrate

Physical vapour deposition



Chemical vapour deposition



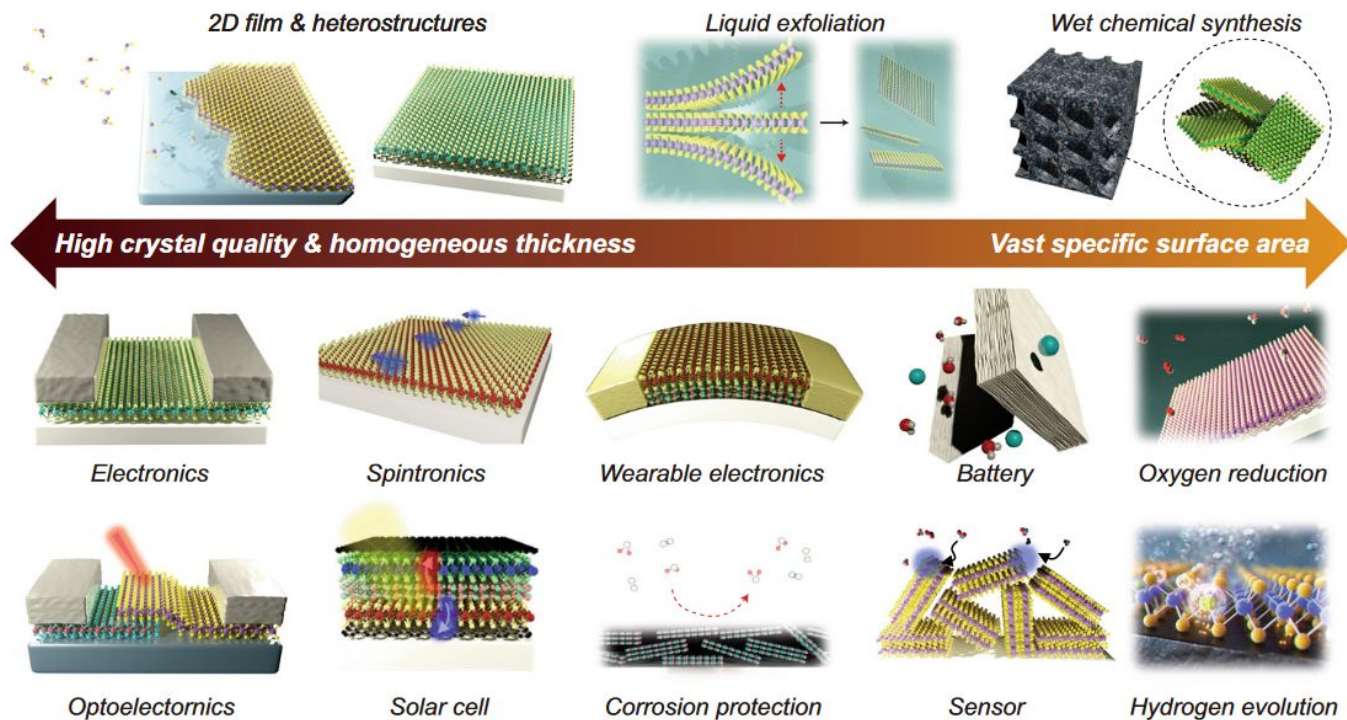
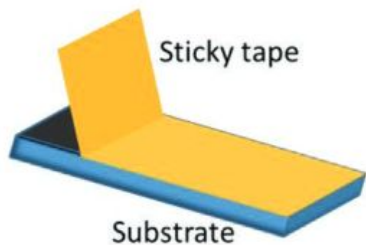
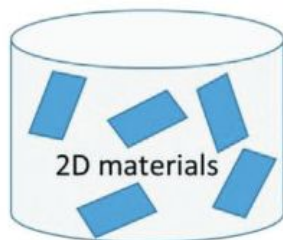


Fig. 1 Specific target-oriented techniques for the mass production of 2D materials. 2D films and heterostructures require high crystal quality and homogeneous thickness for applications such as electronics and spintronics, whereas high-porosity powders with vast specific surface area can be used in contexts such as catalysts and energy storage.

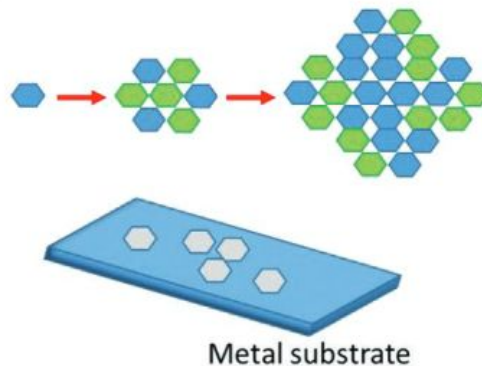
Mechanical cleavage



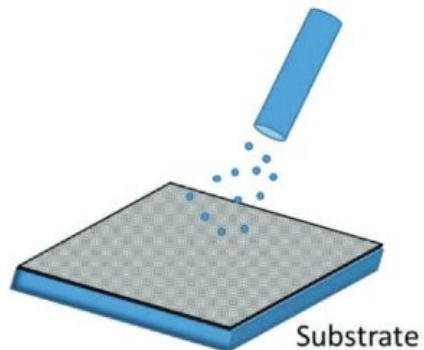
Liquid vapour exfoliation



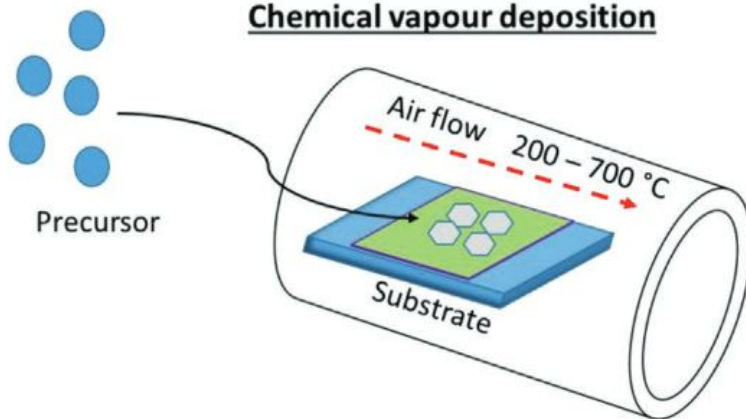
Chemical synthesis



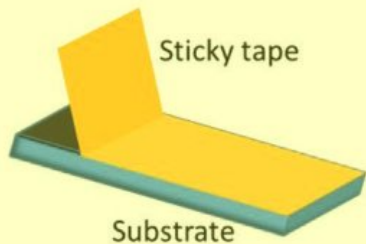
Physical vapour deposition



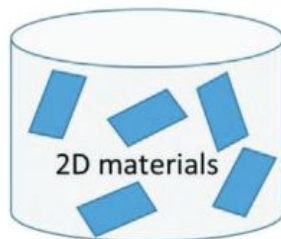
Chemical vapour deposition



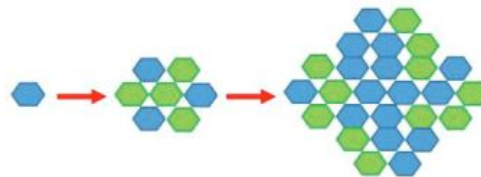
Mechanical cleavage



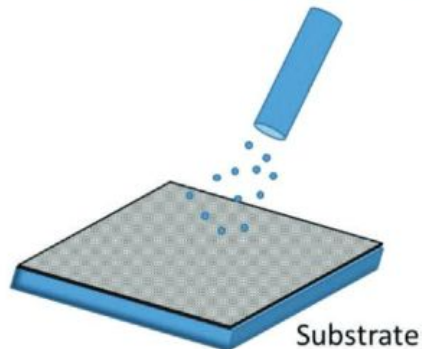
Liquid vapour exfoliation



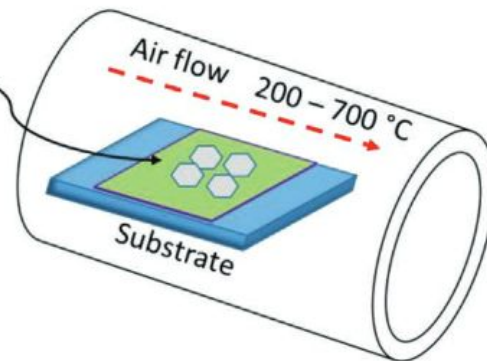
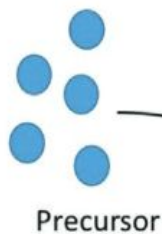
Chemical synthesis



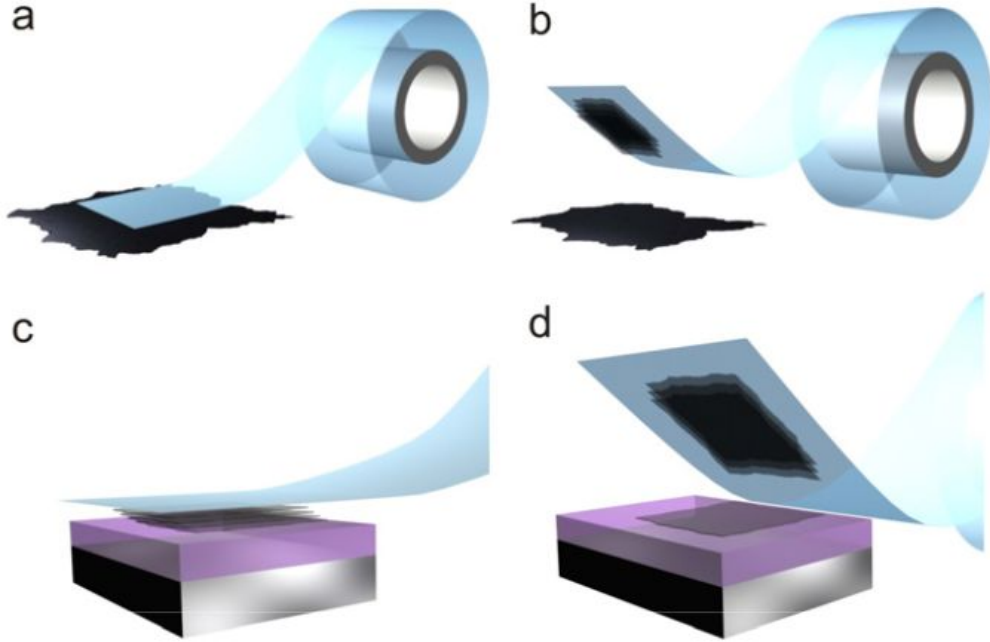
Physical vapour deposition



Chemical vapour deposition



Micromechanical cleavage



Physics
2010

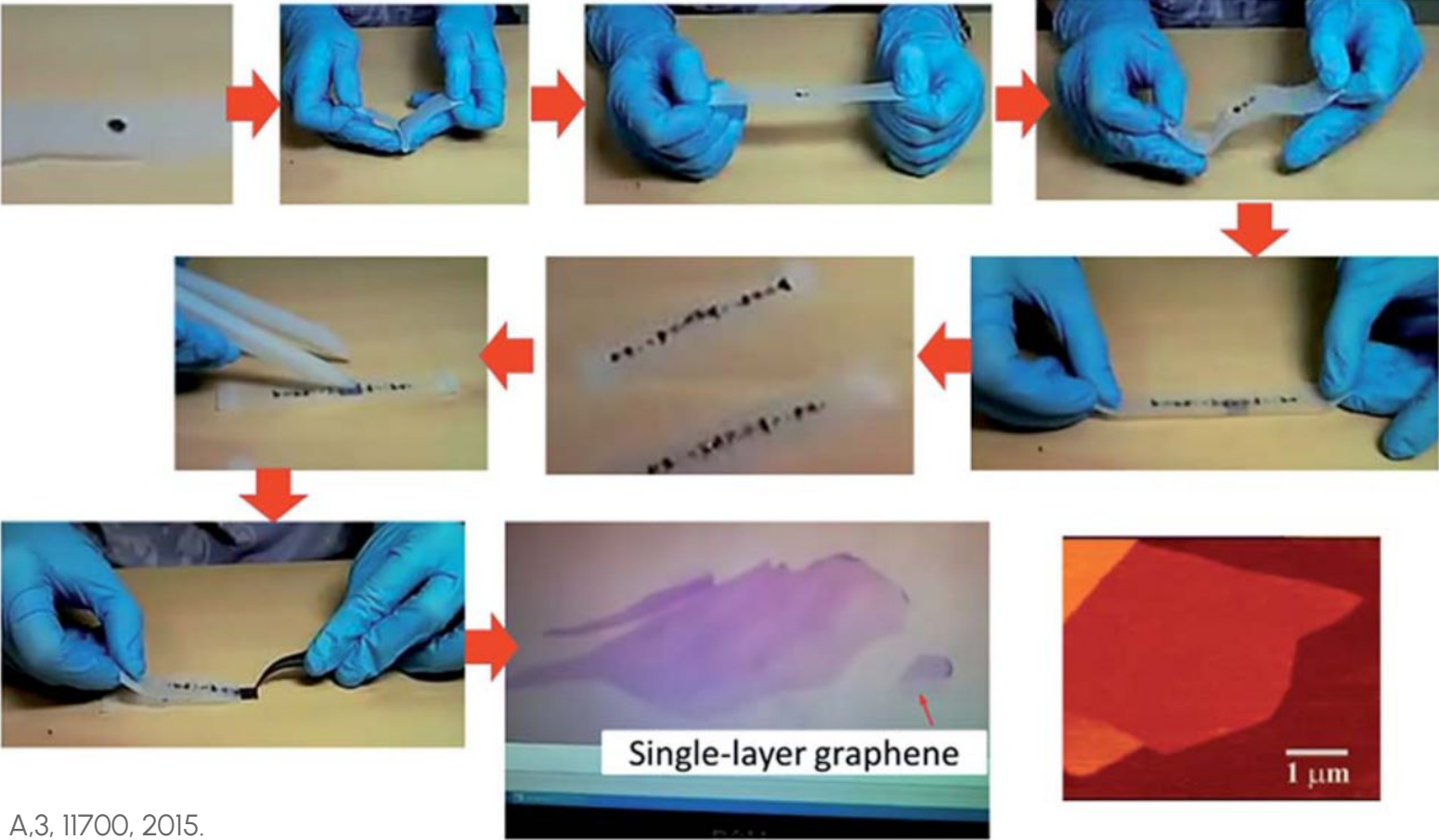
Electric Field Effect in Atomically Thin Carbon Films

K. S. Novoselov,¹ A. K. Geim,^{1*} S. V. Morozov,² D. Jiang,¹
Y. Zhang,¹ S. V. Dubonos,² I. V. Grigorieva,¹ A. A. Firsov²

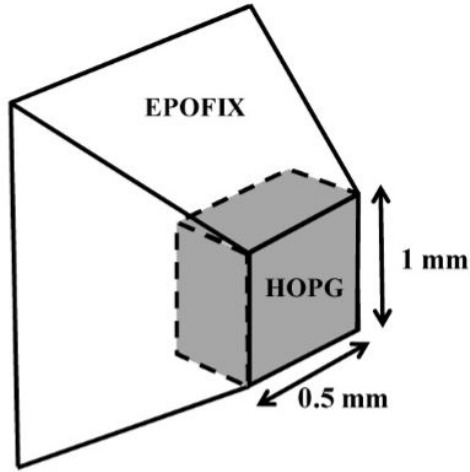
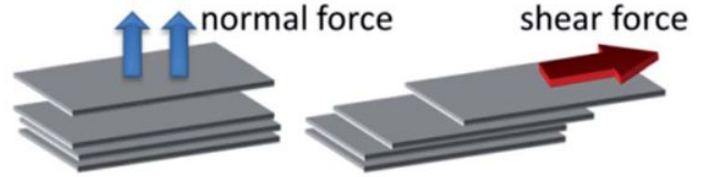
We describe monocrystalline graphitic films, which are a few atoms thick but are nonetheless stable under ambient conditions, metallic, and of remarkably high quality. The films are found to be a two-dimensional semimetal with a tiny overlap



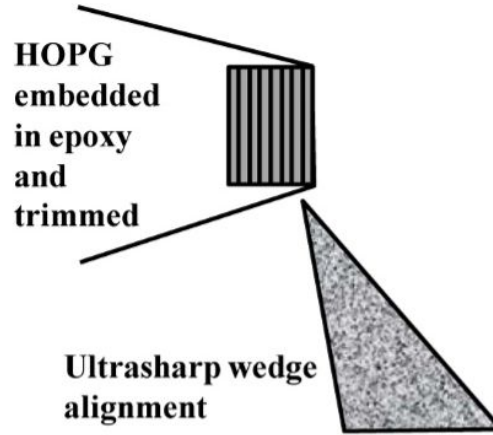
Micromechanical cleavage



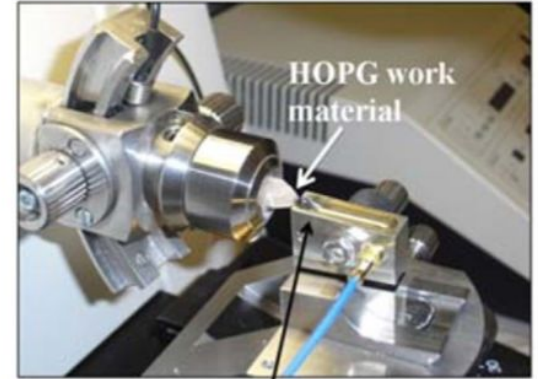
Micromechanical cleavage



(a)



(b)



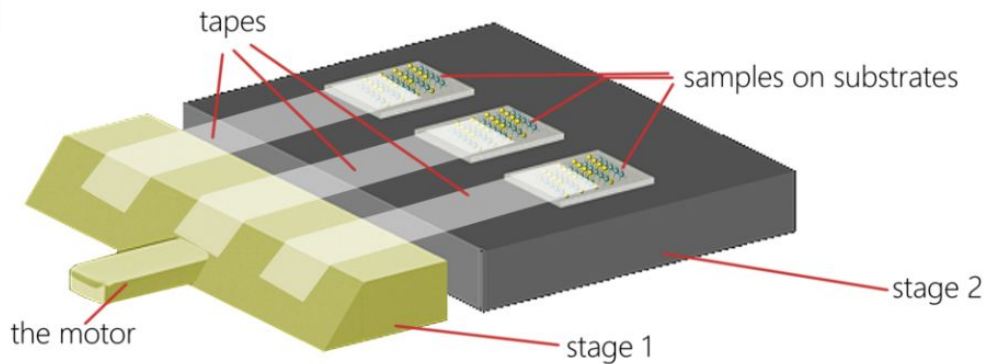
Ultra sharp wedge mounted on an ultrasonic oscillation system

(c)

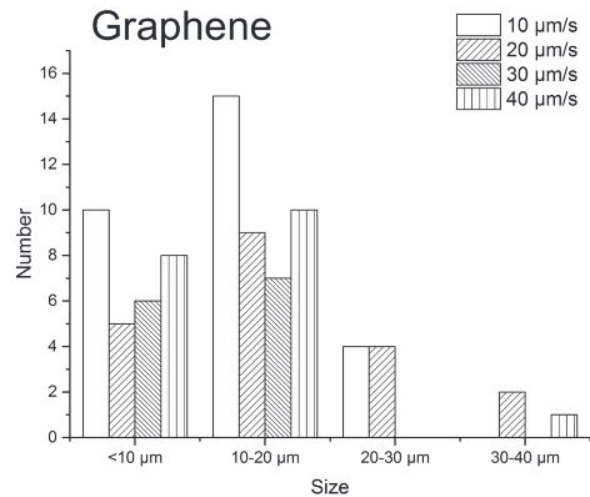
Figure 1 HOPG, SPI grade ZYH. (a) HOPG mounted in epofix and trimmed to pyramid shape. (b) Setup showing wedge alignment with HOPG layers. (c) Actual experimental setup.

Micromechanical cleavage

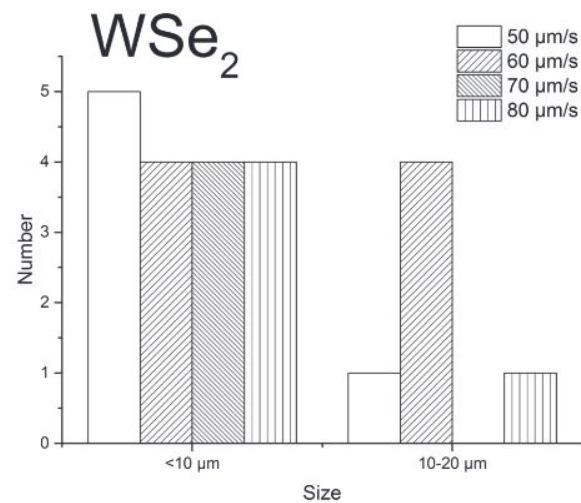
(b)



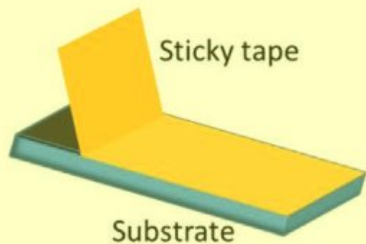
(a)



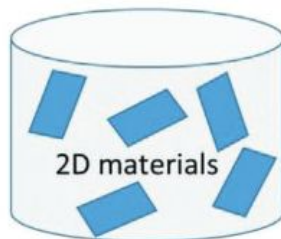
(b)



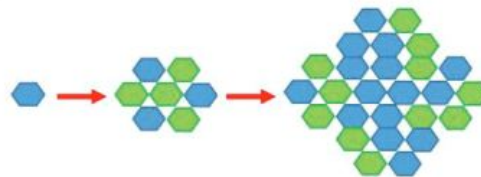
Mechanical cleavage



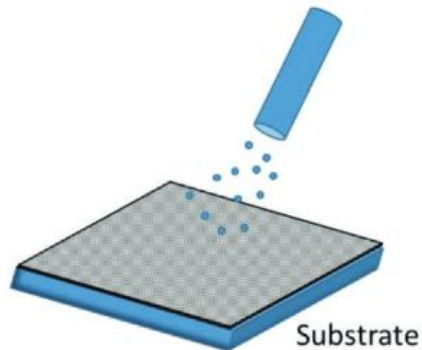
Liquid vapour exfoliation



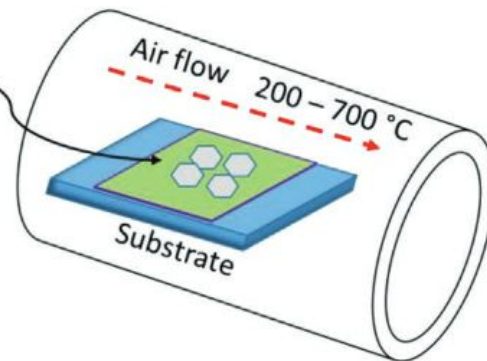
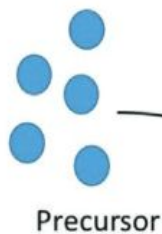
Chemical synthesis



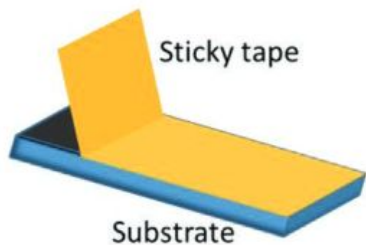
Physical vapour deposition



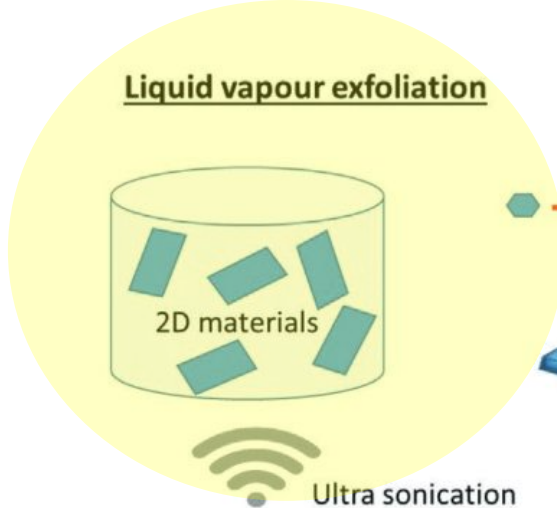
Chemical vapour deposition



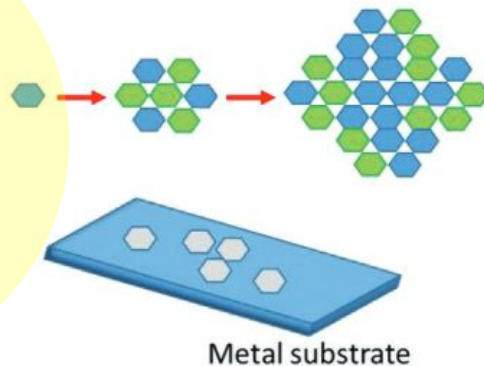
Mechanical cleavage



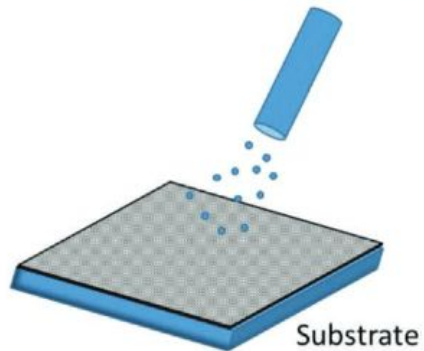
Liquid vapour exfoliation



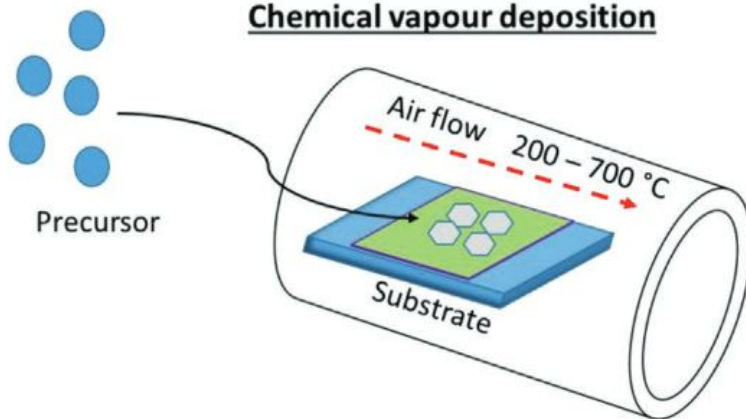
Chemical synthesis



Physical vapour deposition



Chemical vapour deposition



High-yield production of graphene by liquid-phase exfoliation of graphite

YENNY HERNANDEZ^{1†}, VALERIA NICOLOSI^{1†}, MUSTAFA LOTYA¹, FIONA M. BLIGHE¹, ZHENYU SUN^{1,2},
SUKANTA DE^{1,2}, I. T. McGOVERN¹, BRENDAN HOLLAND¹, MICHELE BYRNE³, YURII K. GUN'KO^{2,3},
JOHN J. BOLAND^{2,3}, PETER NIRAJ^{2,3}, GEORG DUESBERG^{2,3}, SATHEESH KRISHNAMURTHY^{2,3},
ROBBIE GOODHUE⁴, JOHN HUTCHISON⁵, VITTORIO SCARDACI⁶, ANDREA C. FERRARI⁶
AND JONATHAN N. COLEMAN^{1,2*}

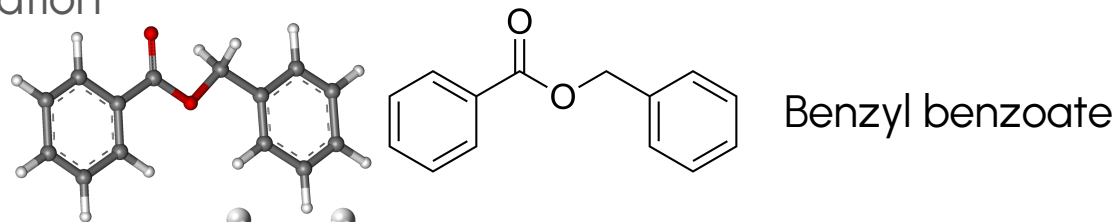
¹School of Physics, Trinity College Dublin, Dublin 2, Ireland

²Centre for Research on Adaptive Nanostructures and Nanodevices (CRANN), Trinity College Dublin, Dublin 2, Ireland

³School of Chemistry, Trinity College Dublin, Dublin 2, Ireland

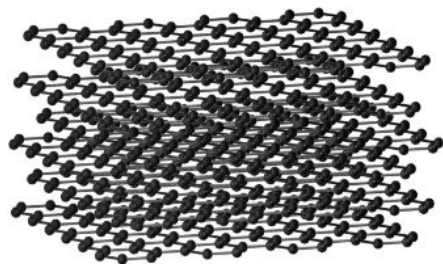
Published online: 10 August 2008; doi:10.1038/nnano.2008.215

Liquid Phase Exfoliation

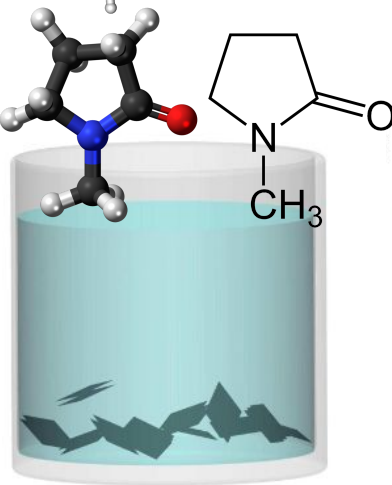


Benzyl benzoate

N-methylpyrrolidone (NMP)



Starting material



Dispersion
in solvents

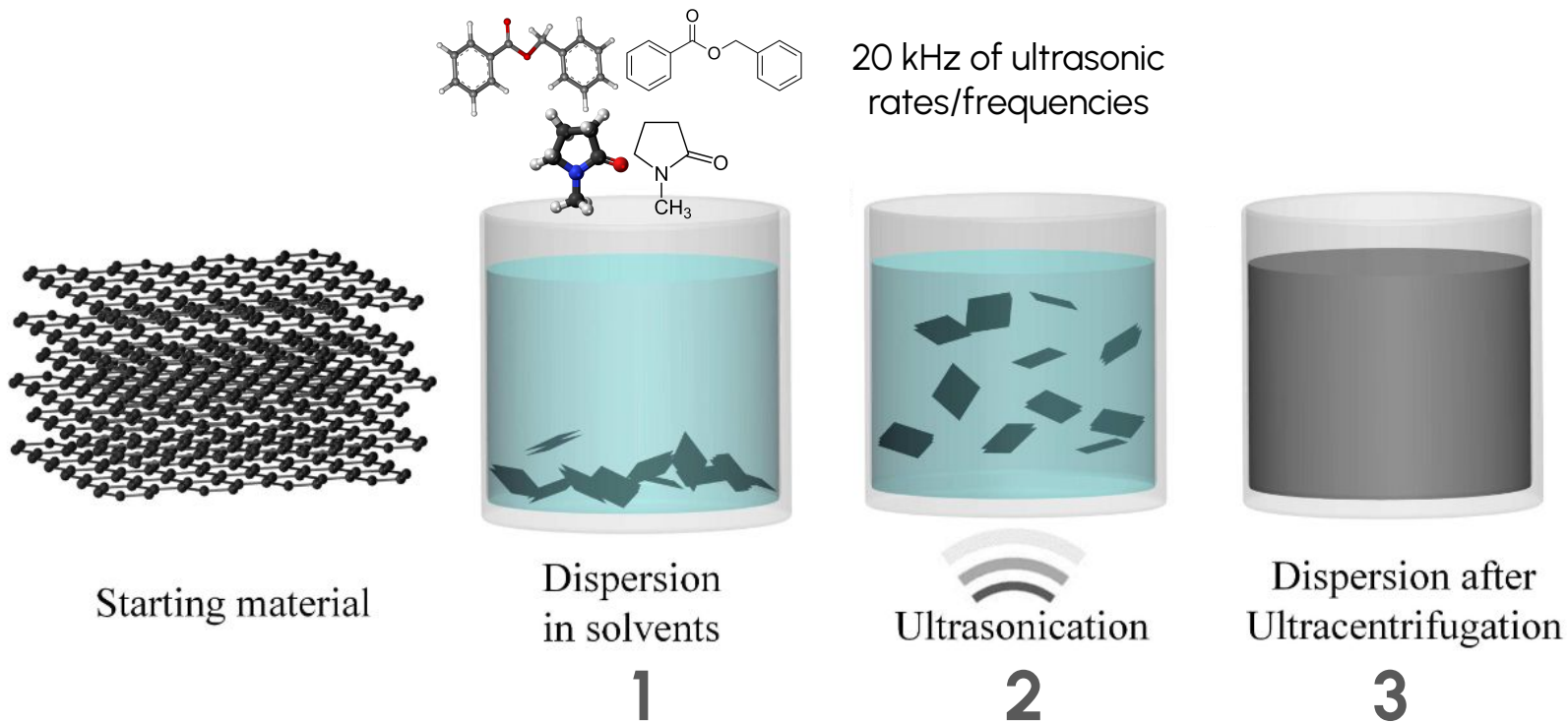


Ultrasonication



Dispersion after
Ultracentrifugation

Liquid Phase Exfoliation



Liquid Phase Exfoliation

Esfoliação só ocorre se o custo energético é pequeno

$$\frac{\Delta H_{\text{mix}}}{V_{\text{mix}}} \approx \frac{2}{T_{\text{flake}}} (\delta_{\text{G}} - \delta_{\text{sol}})^2 \phi$$



$\delta_i = \sqrt{(E_{\text{sur}}^i)}$ is the square root of the surface energy of phase i ,

$$\delta_{\text{Graphite}} \sim 70 - 80 \text{ mJ m}^{-2}$$

Liquid Phase Exfoliation of non-vdW materials: MX-enes

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Two-Dimensional Nanocrystals Produced by Exfoliation of Ti_3AlC_2

Michael Naguib, Murat Kurtoglu, Volker Presser, Jun Lu, Junjie Niu, Min Heon, Lars Hultman, Yury Gogotsi,* and Michel W. Barsoum*

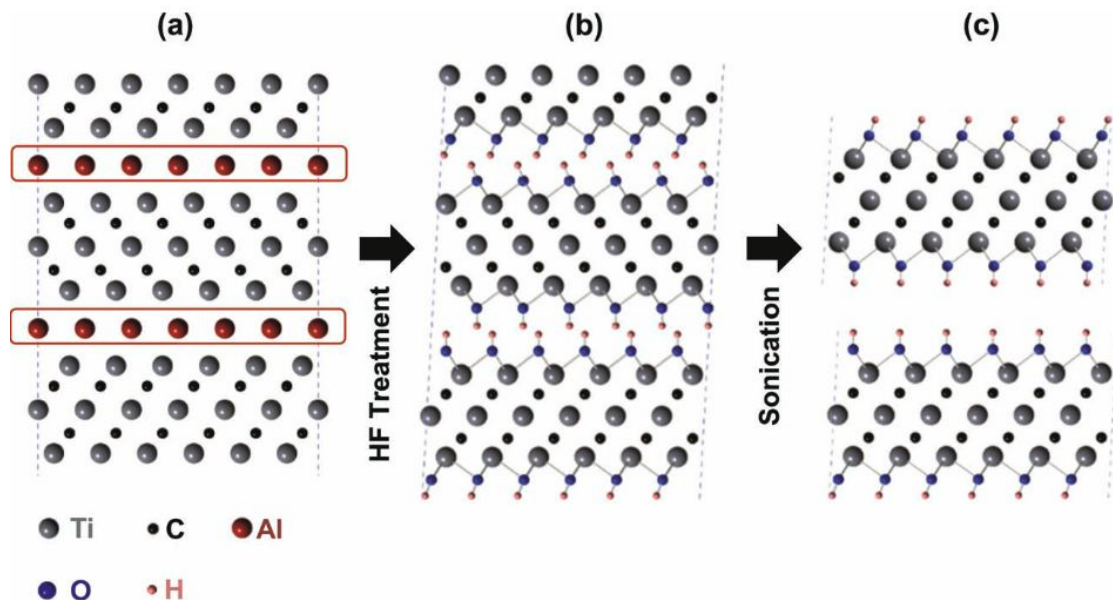
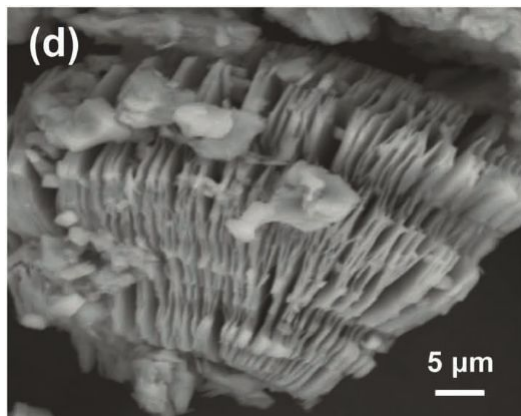


Figure 1. Schematic of the exfoliation process for Ti_3AlC_2 . a) Ti_3AlC_2 structure. b) Al atoms replaced by OH after reaction with HF. c) Breakage of the hydrogen bonds and separation of nanosheets after sonication in methanol.

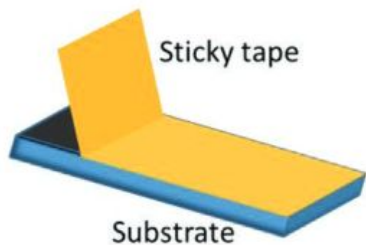
Name Lattice System Year of exfoliation	Crystal Structure	TEM image	HR-TEM/ STEM image
Tungsten oxide (WO ₃) Monoclinic 2017			
Lithium Manganese dioxide (LiMn ₂ O ₄) Cubic 2017			
Boron (β-B) Rhombohedral 2018			
Hematite (α-Fe ₂ O ₃) Rhombohedral 2018			
Titanium (α-Ti) Hexagonal 2019			
Magnesium (Mg) Hexagonal 2019			

Liquid-Phase Exfoliation of Nonlayered Non-Van-Der-Waals Crystals into Nanoplatelets

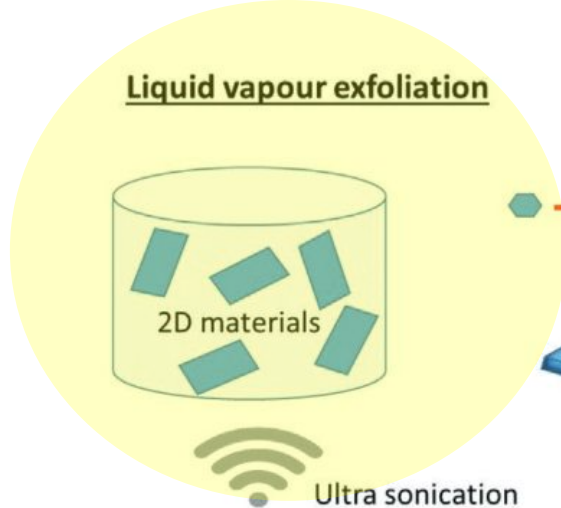
Harnet Kaur and Jonathan N. Coleman*

Pyrite (FeS ₂) Cubic 2020			
Silicon (Si) Cubic 2020			
Germanium (Ge) Cubic 2021			
Tin (Sn) Tetragonal 2021			
Titanium carbide (TiC) Cubic 2021			

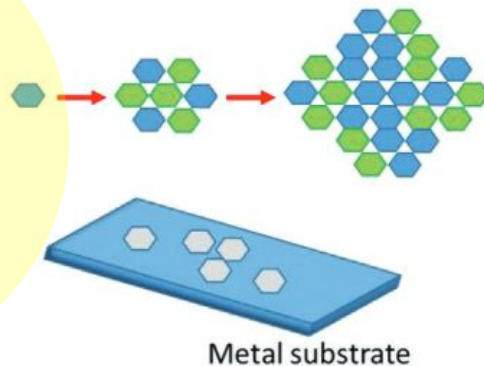
Mechanical cleavage



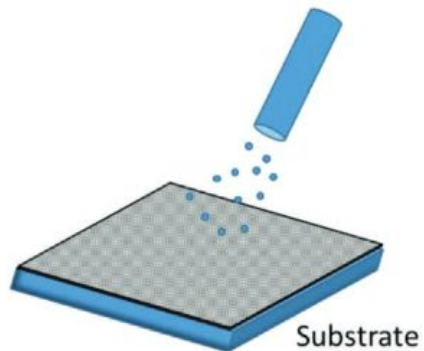
Liquid vapour exfoliation



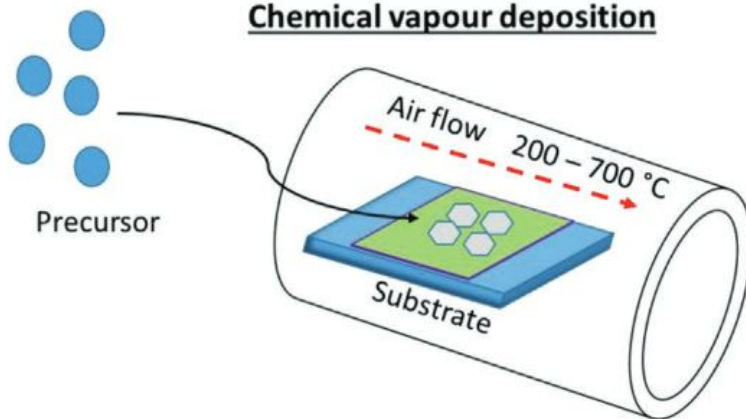
Chemical synthesis



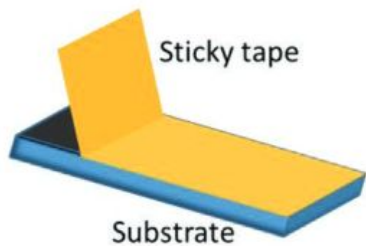
Physical vapour deposition



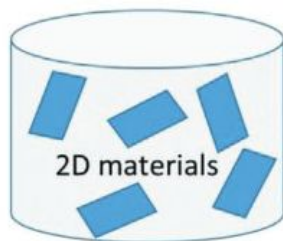
Chemical vapour deposition



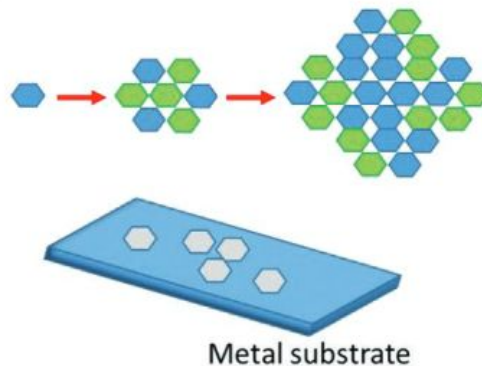
Mechanical cleavage



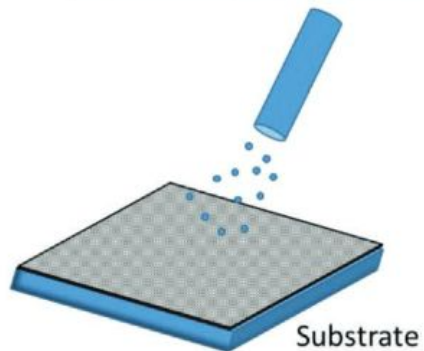
Liquid vapour exfoliation



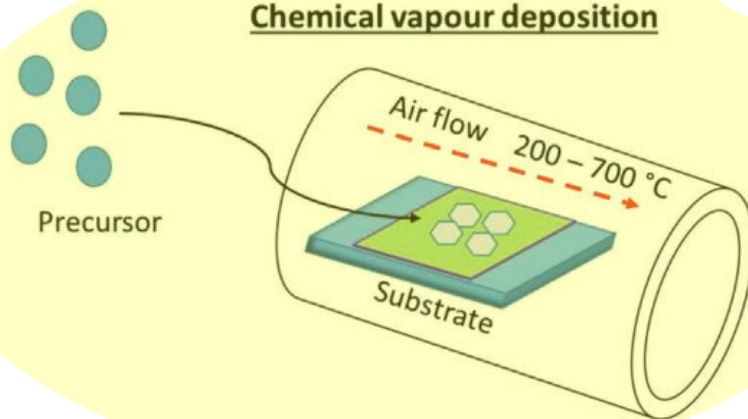
Chemical synthesis



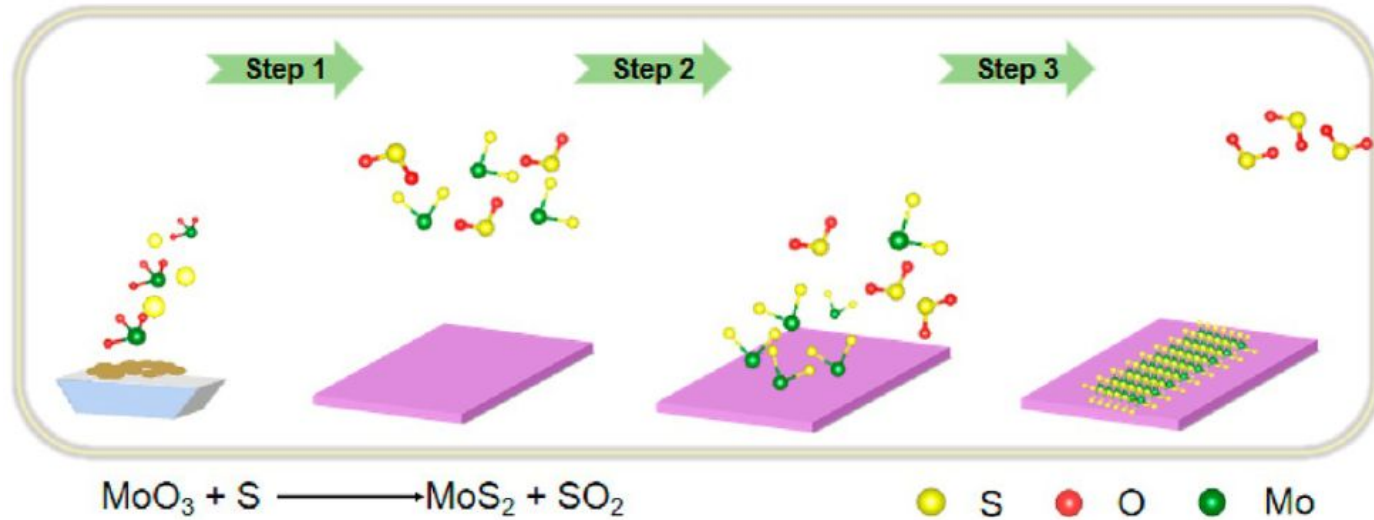
Physical vapour deposition



Chemical vapour deposition



Chemical Vapor Deposition



Temperatura
Pressão
Precursor
Carrier Gas
Substrato

Chemical Vapor Deposition

- Classified by **operating conditions**:
 - **Atmospheric pressure CVD (APCVD)** – CVD at atmospheric pressure.
 - **Low-pressure CVD (LPCVD)** – CVD at sub-atmospheric pressures. Reduced pressures tend to reduce unwanted gas-phase reactions and improve film uniformity across the wafer.
 - **Ultra High vacuum CVD (UHVCVD)** – CVD at very low pressure, typically below 10^{-6} Pa.
 - **Sub-atmospheric CVD (SACVD)** – CVD at sub-atmospheric pressures.
- Classified by **physical characteristics of vapor**:
 - **Aerosol assisted CVD (AACVD)** – the precursors are transported to the substrate by means of a liquid/gas aerosol, which can be generated ultrasonically. Suitable for use with non-volatile precursors.
 - **Direct liquid injection CVD (DLICVD)** – the precursors are in liquid form (liquid or solid dissolved in a convenient solvent). Liquid solutions are injected in a vaporization chamber towards injectors (typically car injectors).
 - **Metal organic CVD (MOCVD)** – based on metal organic precursors. Utilizes chemical compounds with low to moderate vapor pressure as precursors.
- Classified by **type of substrate heating**:
 - **Hot wall CVD** – chamber is heated by an external power source and the substrate is heated by radiation from the heated chamber walls.
 - **Cold wall CVD** – only the substrate is directly heated either by induction or by passing current through the substrate itself.

Chemical Vapor Deposition: Important parameters

Temperature: one of the main parameters (correlated to the free energy of the reactants). Direct impact on the uniformity and composition. Can promote uniform deposition (higher quality). At high $T (> T_{\text{growth}})$ → single crystal, polycrystalline, amorphous or not even deposition.

Pressure: usually, $P = P_{\text{atm}}$. However, low P can help uniformity.

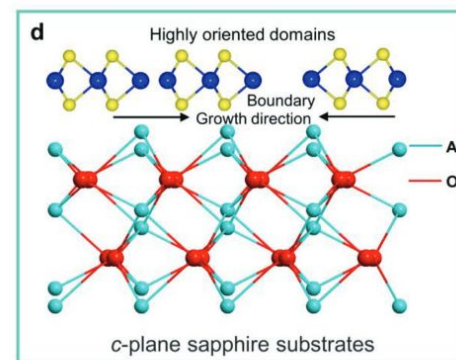
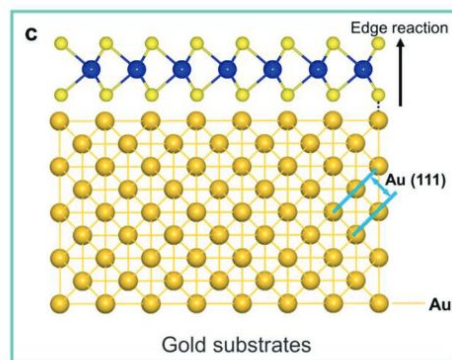
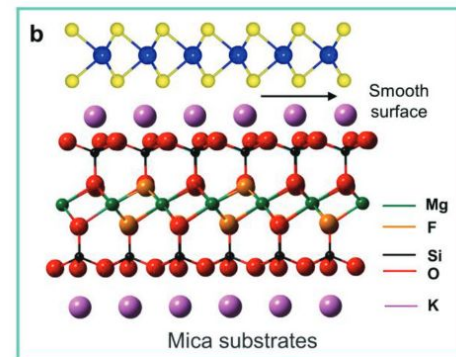
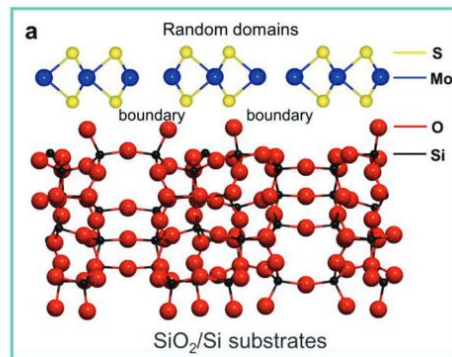
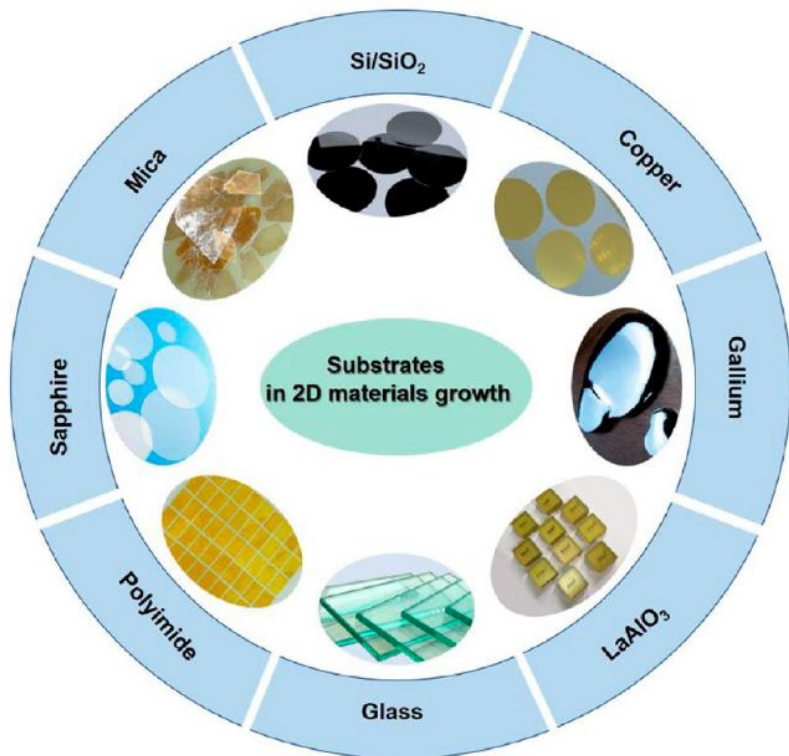
Precursor: gas and solid are common precursors (although gaseous can be better controlled). Relative proportions are important. In recent years, salts (e.g. NaCl) have been added to generate a transition product ($\text{MoO}_3 + \text{NaCl} \rightarrow \text{MoO}_2\text{Cl}_2$) and improve growth rate and crystallinity.

Carrier Gas: rate, ratio, and supply sequence of the carrier gas can affect the size, nucleation density, morphology, and thickness. Reducing the gas flow → reduction of the nucleation density → large-size single crystals with few grain boundaries.

Substrate

Choosing the substrate

Strong impact on the number of nucleation points, size, and quality of the material.



Choosing the substrate

pubs.acs.org/cm

Review

Substrates in the Synthesis of Two-Dimensional Materials via Chemical Vapor Deposition

Biao Qin,^{||} Huifang Ma,^{||} Mongur Hossain, Mianzeng Zhong, Qinglin Xia, Bo Li,^{*} and Xidong Duan^{*}



Cite This: <https://dx.doi.org/10.1021/acs.chemmater.0c03549>



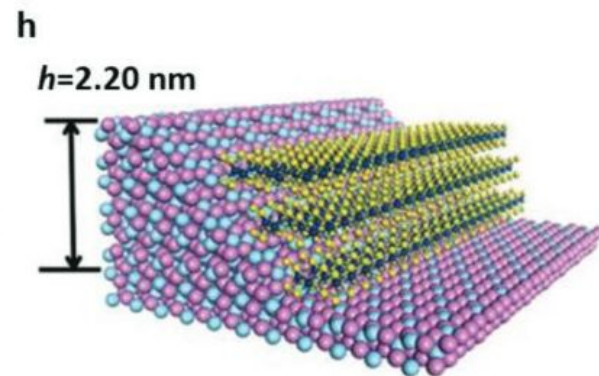
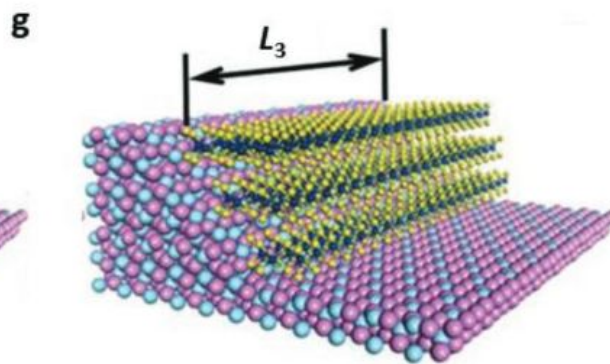
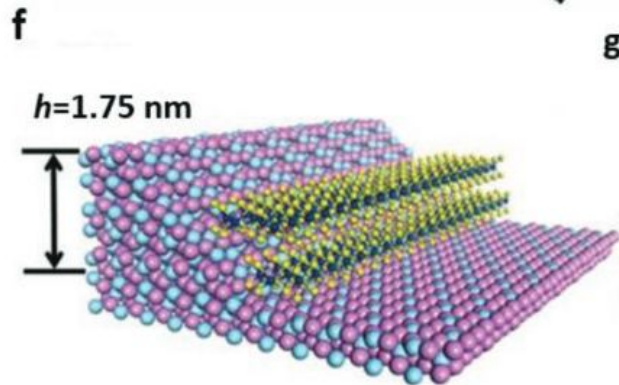
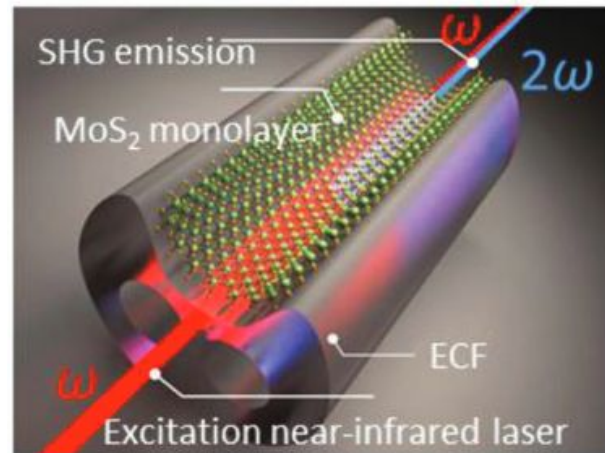
Read Online

*"Substrates play three important roles in the synthesis of 2DMs by the CVD method:
adsorbing source materials,
promoting material nucleation,
and stimulating material epitaxial growth."*

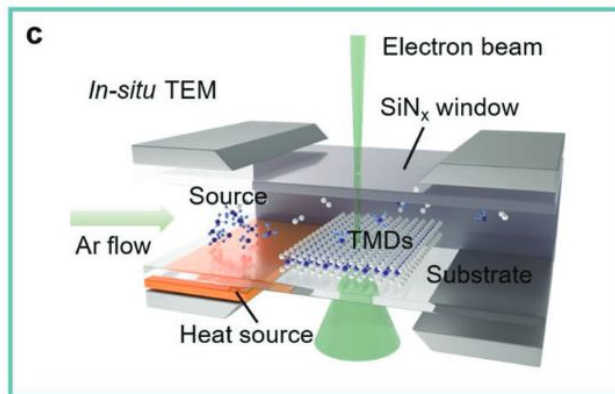
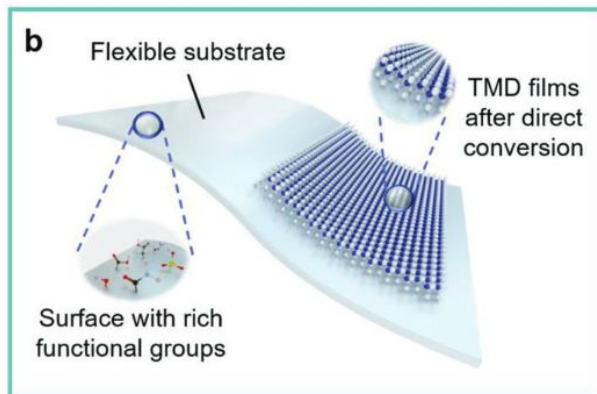
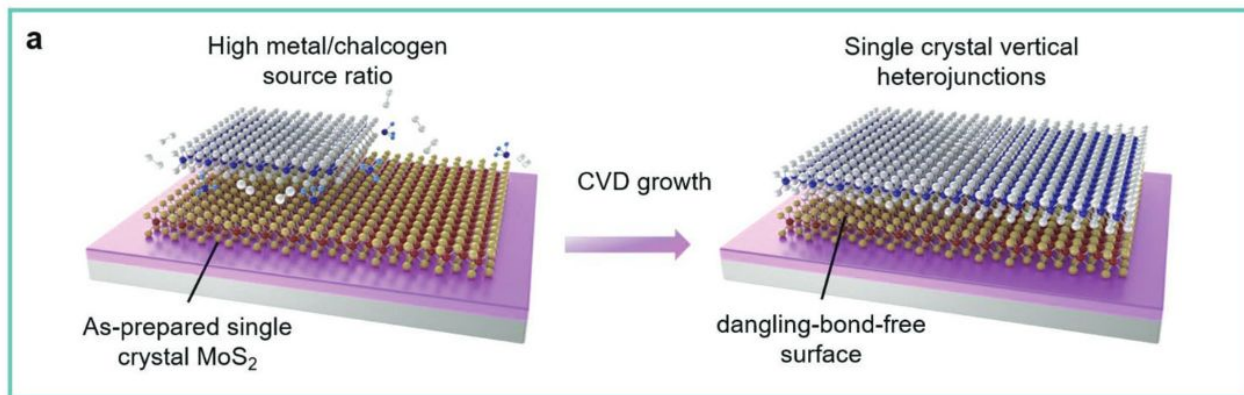
*"...key factors are
the **matching degree of the lattice constant** of the target product and the substrate,
the **thermal expansion coefficient**,
and the **catalytic effect** of the specific substrate on the material."*



Substrate Engineering



Substrate Engineering

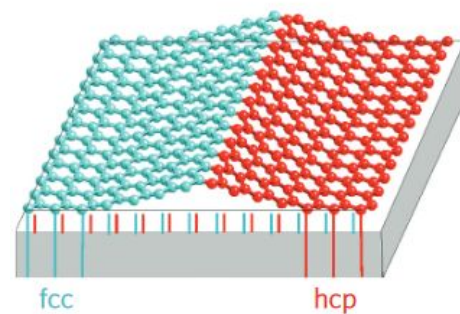
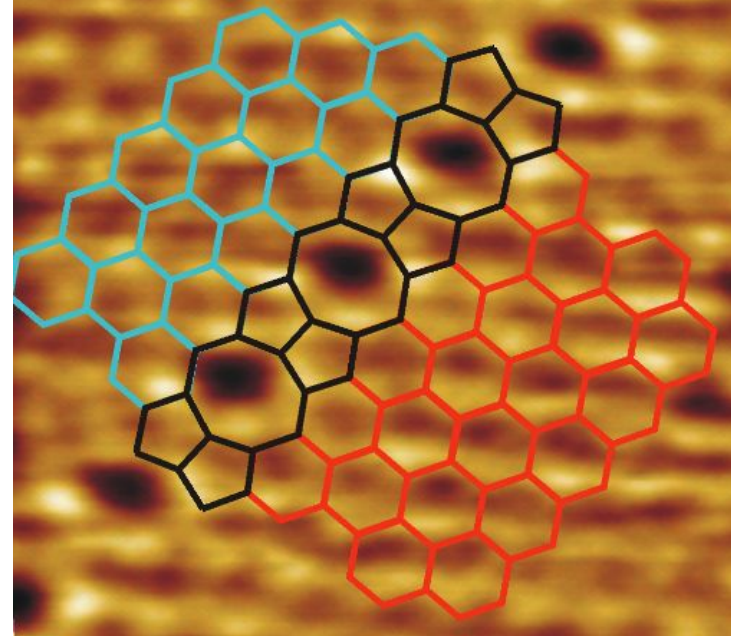
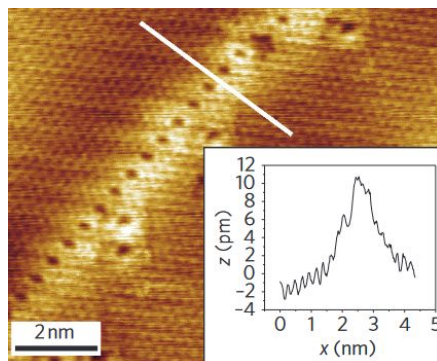
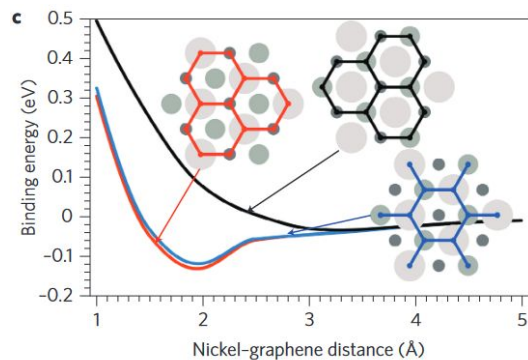
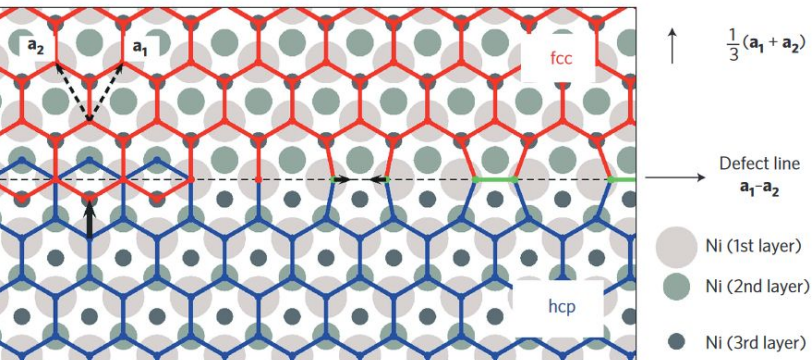


An extended defect in graphene as a metallic wire

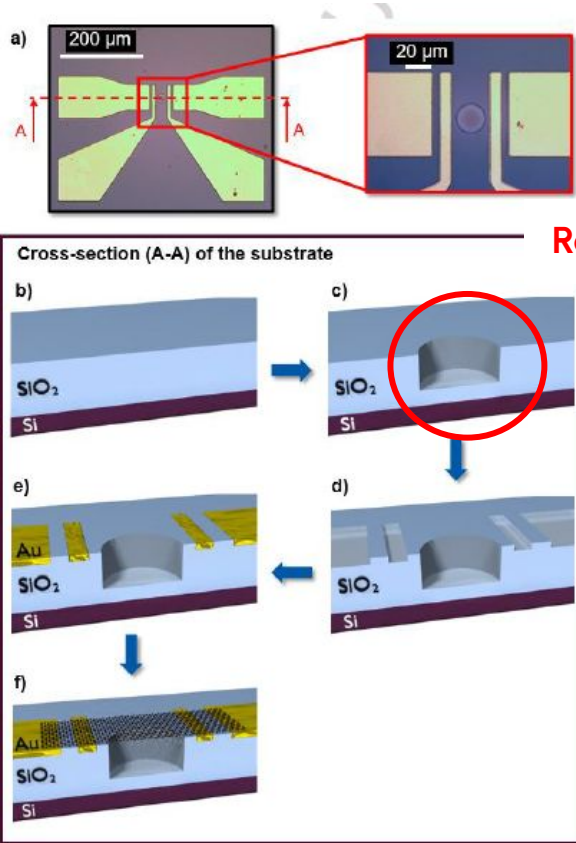
Jayeeta Lahiri, You Lin, Pinar Bozkurt, Ivan I. Oleynik and Matthias Batzill*

Many proposed applications of graphene require the ability to tune its electronic structure at the nanoscale^{1,2}. Although charge transfer³ and field-effect doping⁴ can be applied to manipulate charge carrier concentrations, using them to

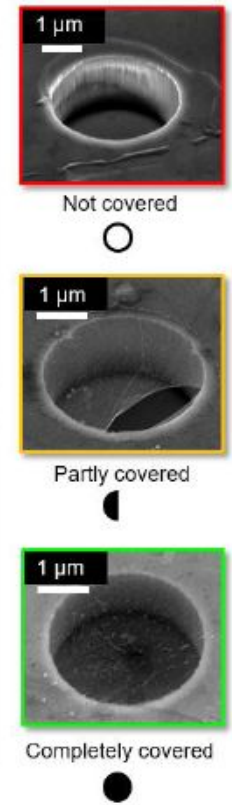
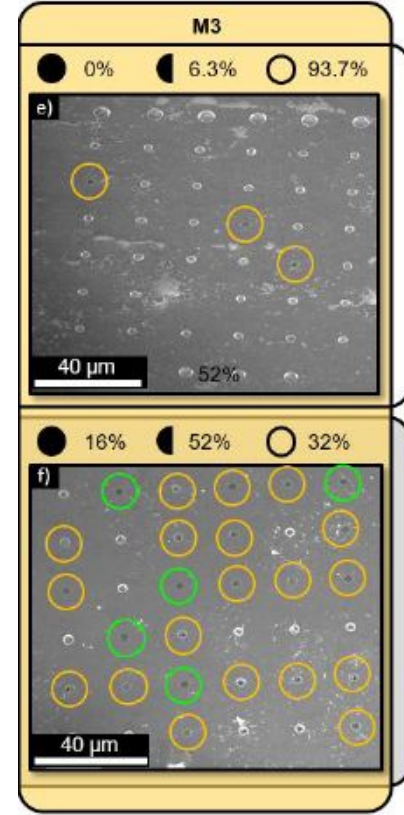
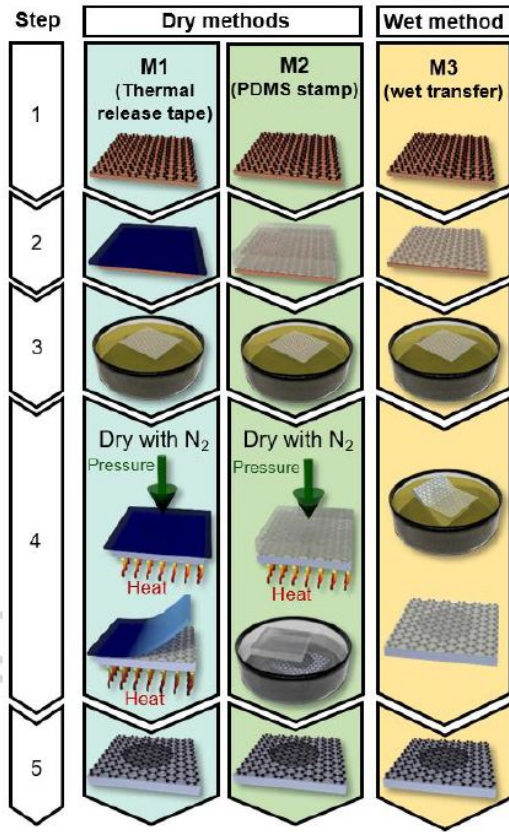
sheets in registry to each other with atomic precision. Such a scaffold can only be a two-dimensional atomic lattice for which graphene has a close epitaxial relationship, such as Ni(111). Graphene grown on Ni(111) with half of the carbon atoms situated



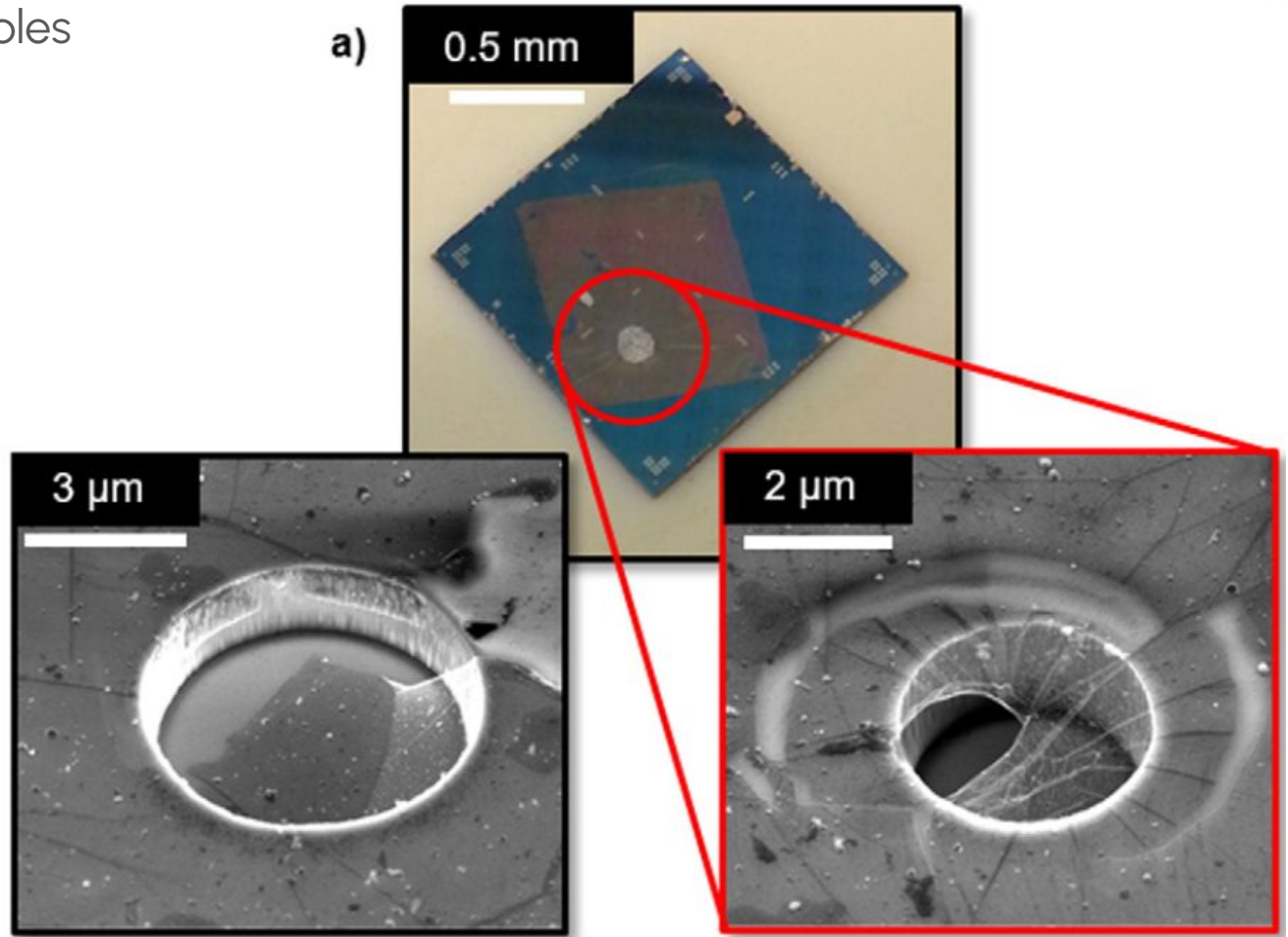
Free-Standing samples



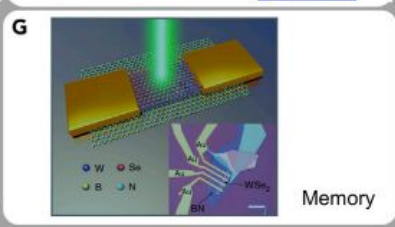
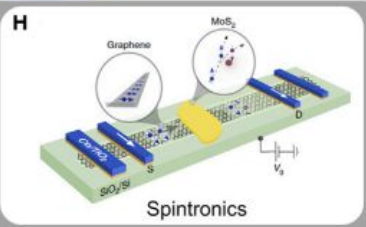
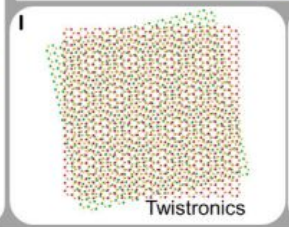
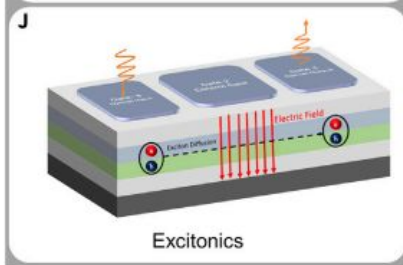
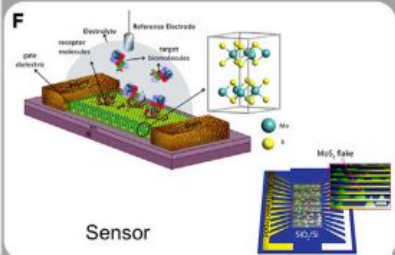
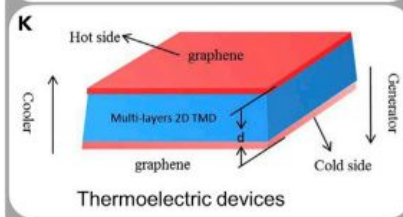
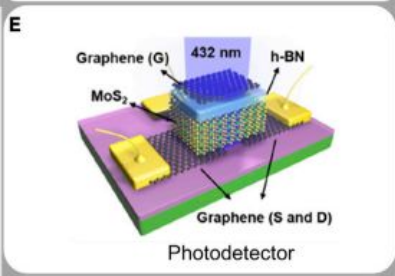
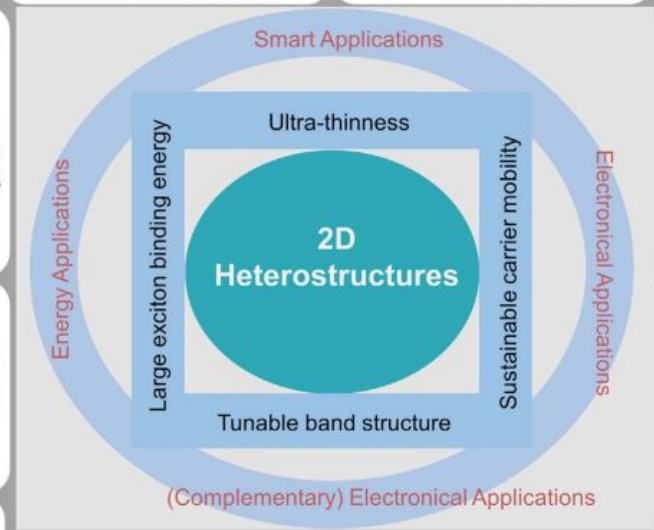
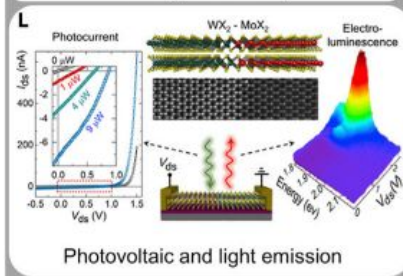
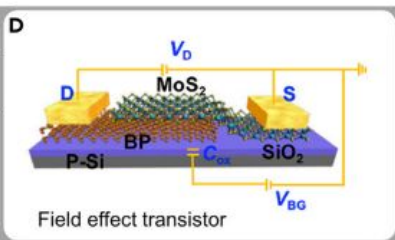
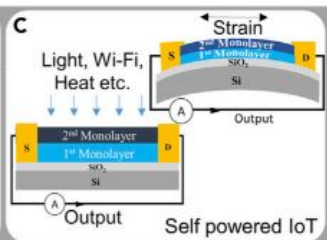
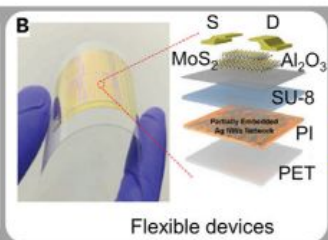
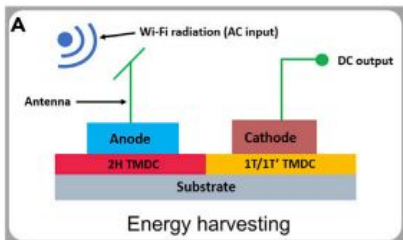
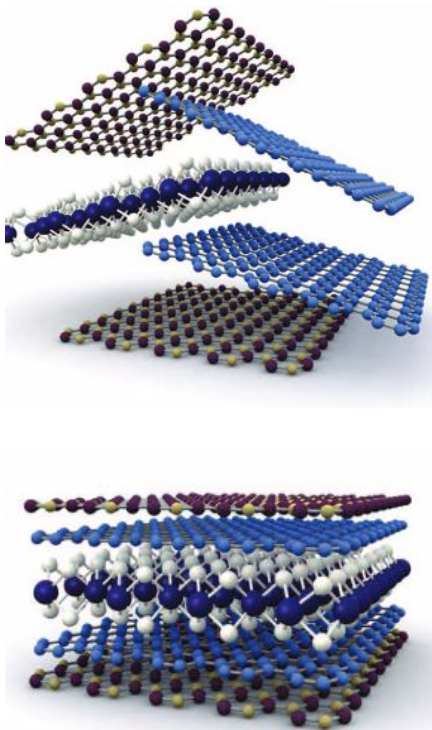
Real



Free-Standing samples

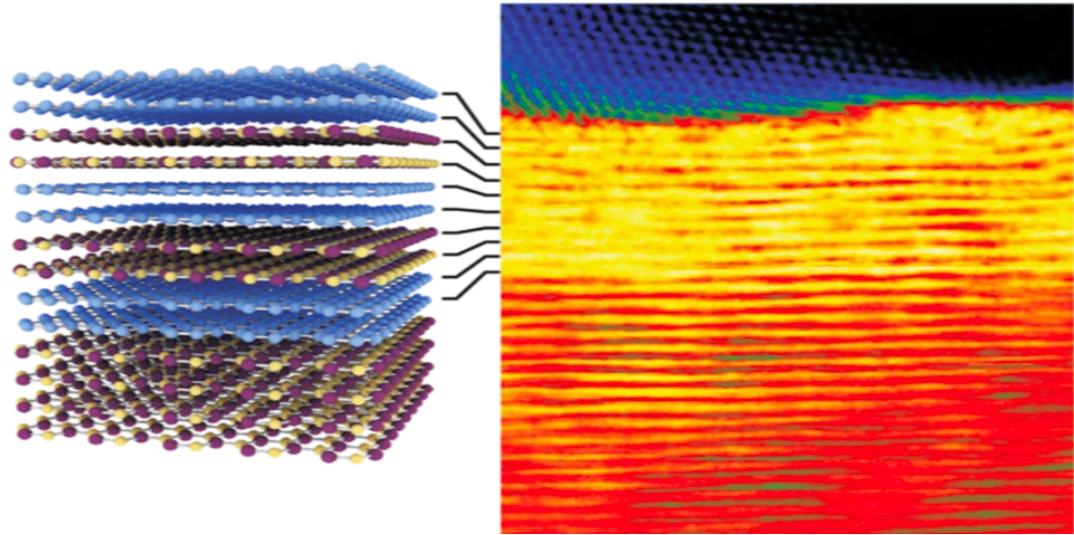
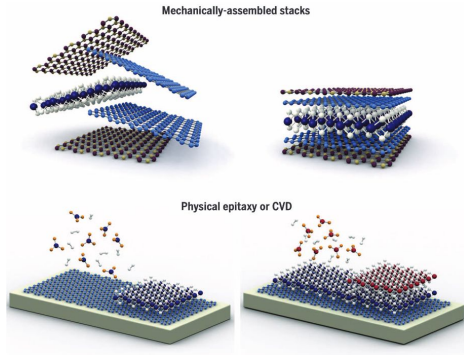


Heteroestructuras



Cross-sectional imaging of individual layers and buried interfaces of graphene-based heterostructures and superlattices

S. J. Haigh^{1*}, A. Gholinia¹, R. Jalil², S. Romani³, L. Britnell², D. C. Elias², K. S. Novoselov²,
L. A. Ponomarenko², A. K. Geim² and R. Gorbachev^{2*}



S. J. Haigh et al. Nature Materials (2012)
Butler et. al. ACS Nano, 7, 2898 (2013)
K. S. Novoselov et. al, Science, 353 (2016)

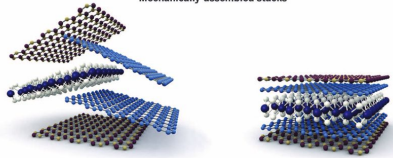
Heteroestructuras

Direct Synthesis of van der Waals Solids

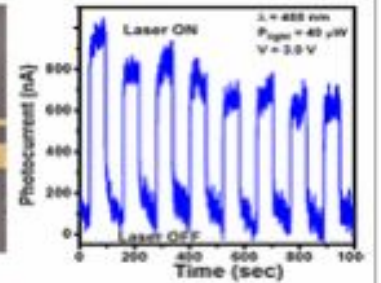
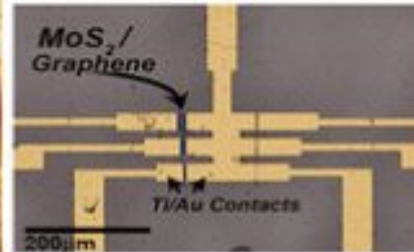
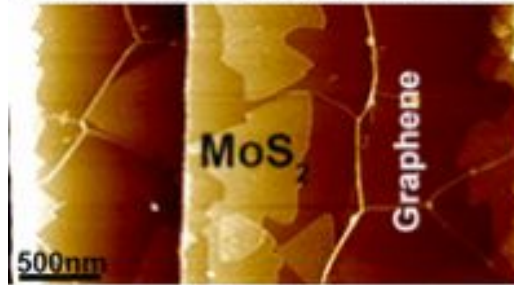
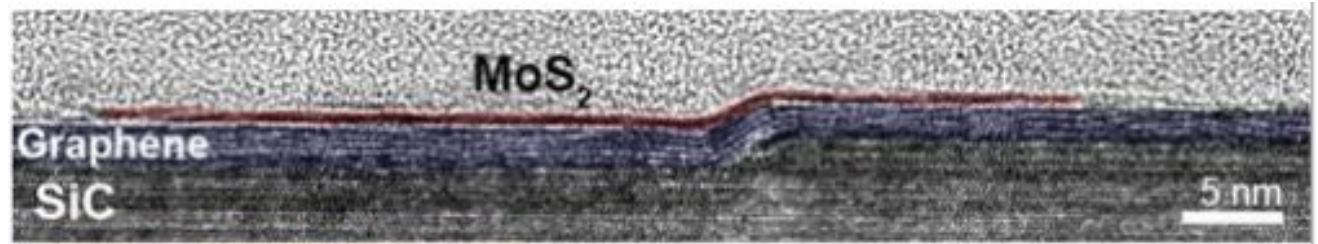
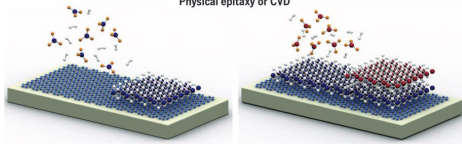
Yu-Chuan Lin,[†] Ning Lu,[‡] Nestor Perea-Lopez,[§] Jie Li,[‡] Zhong Lin,[§] Xin Peng,[‡] Chia Hui Lee,[‡] Ce Sun,[‡] Lazaro Calderin,[†] Paul N. Browning,[†] Michael S. Bresnehan,[†] Moon J. Kim,[‡] Theresa S. Mayer,[‡] Mauricio Terrones,[§] and Joshua A. Robinson^{†,*}

[†]Department of Materials Science and Engineering and Center for 2-Dimensional and Layered Materials, The Pennsylvania State University, University Park, Pennsylvania 16802, United States, [‡]Department of Materials Science and Engineering, The University of Texas at Dallas, Richardson, Texas 75080, United States, [§]Department of Physics and Center for 2-Dimensional and Layered Materials, The Pennsylvania State University, University Park, Pennsylvania 16802, United States, and ^{*}Department of Electrical Engineering, The Pennsylvania State University, State College, University Park, Pennsylvania 16802, United States

Mechanically-assembled stacks



Physical epitaxy or CVD

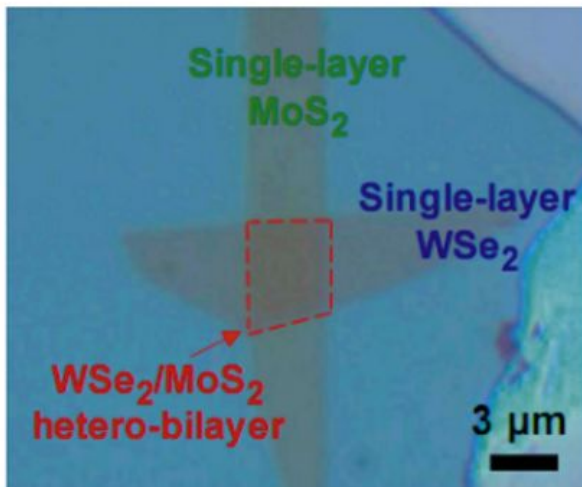
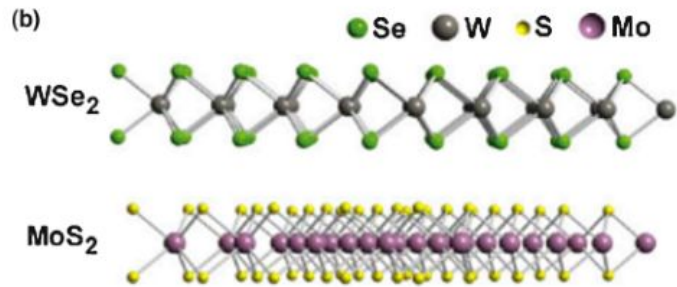


S. J. Haigh et al. Nature Materials (2012)

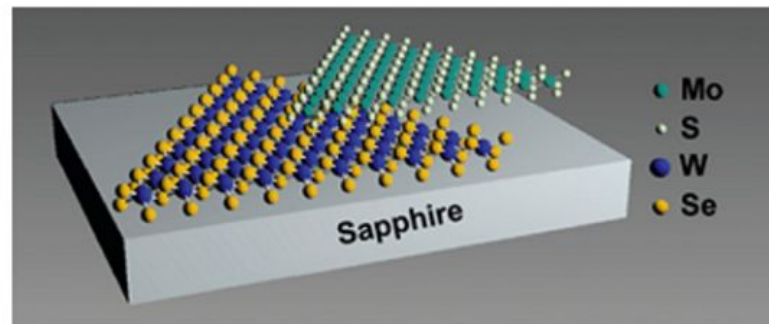
Butler et. al. ACS Nano, 7, 2898 (2013)

K. S. Novoselov et. al, Science, 353 (2016)

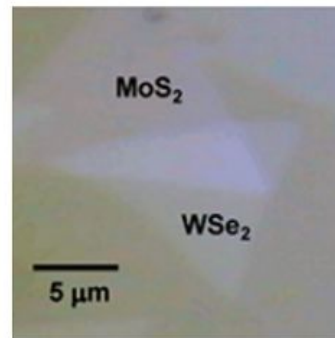
Heteroestructuras



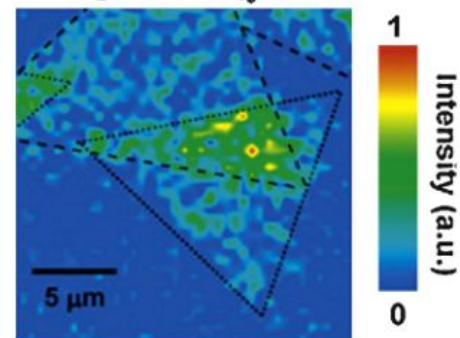
(c)



WSe₂/MoS₂ Heterojunction

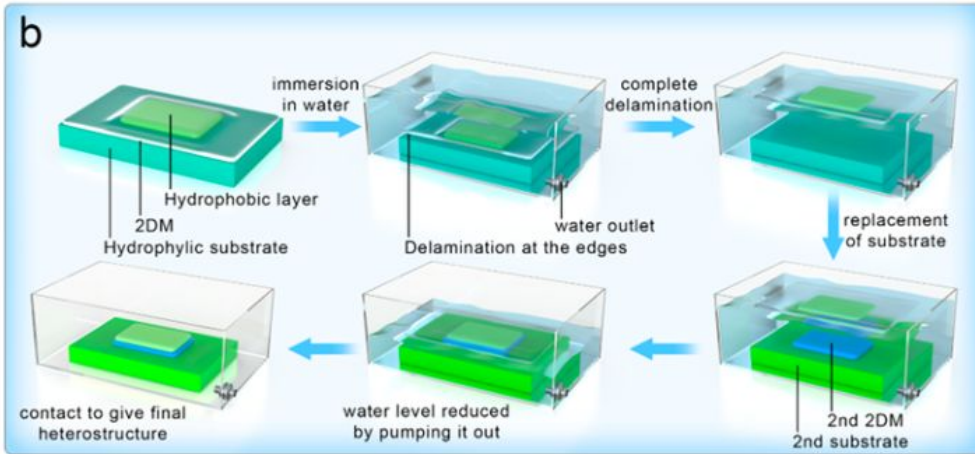
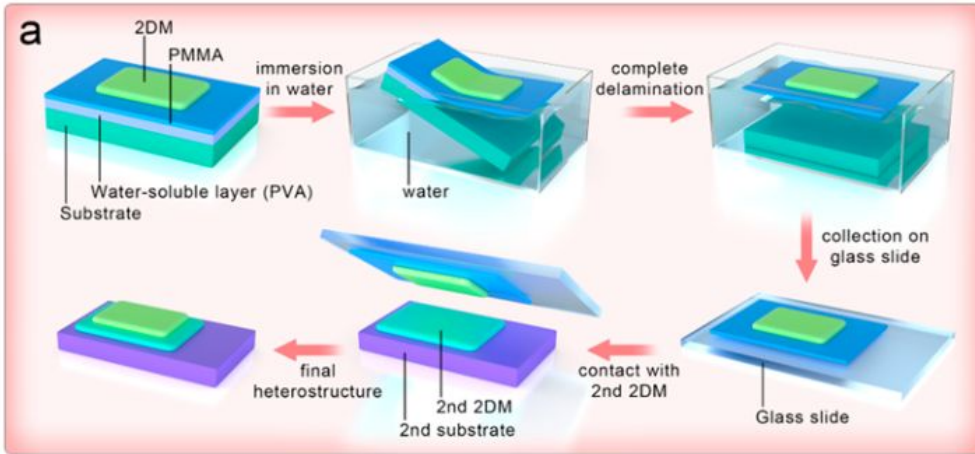


WSe₂ Raman A_{1g}² mode



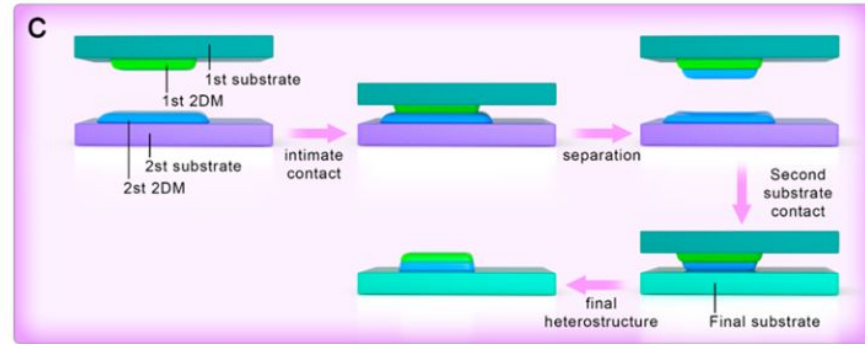
PMMA transfer method

Poly (methyl methacrylate) (PMMA): thermoplastic synthetic polymer



Wedging method

vdW lift and transfer method.



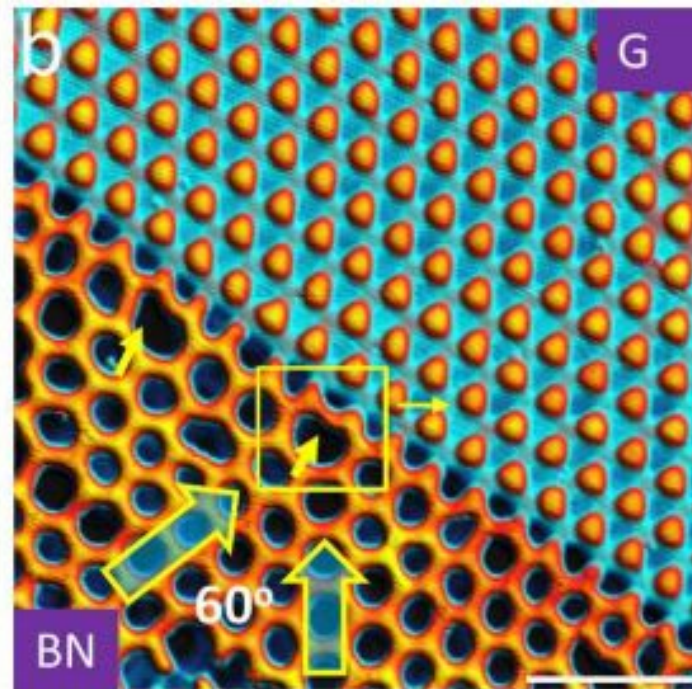
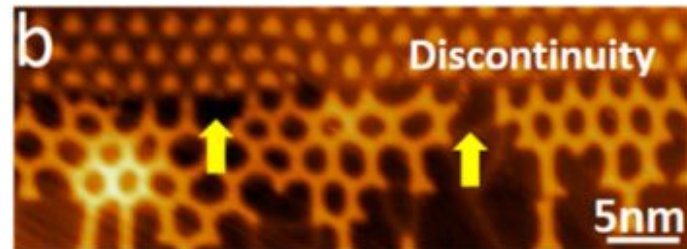
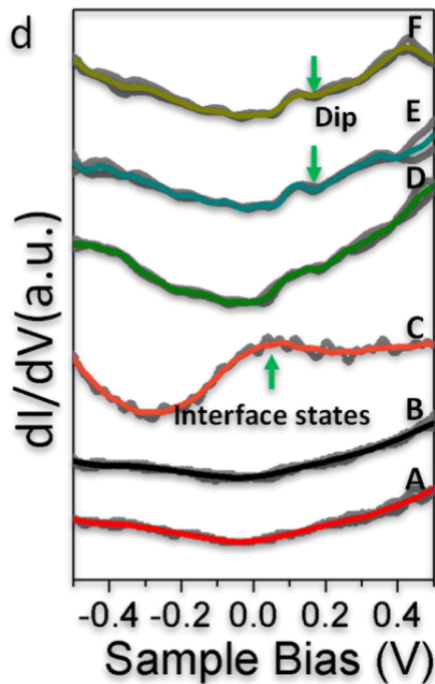
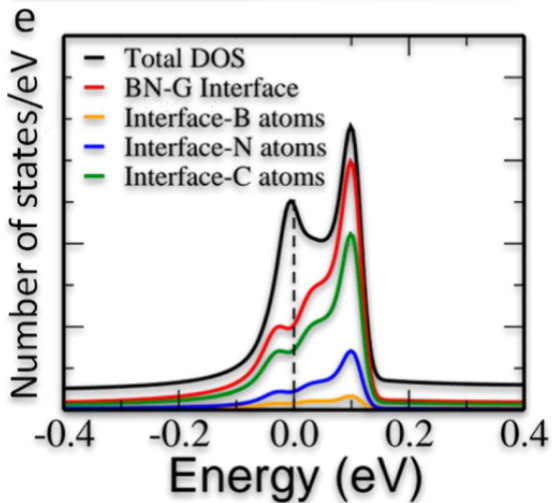
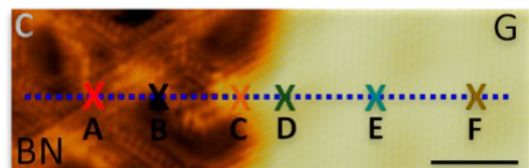
Heteroestructuras

NANO LETTERS

Letter
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Lattice Relaxation at the Interface of Two-Dimensional Crystals: Graphene and Hexagonal Boron-Nitride

Jiong Lu,^{†,‡,¶} Lídia C. Gomes,^{‡,§,#} Ricardo W. Nunes,[§] A. H. Castro Neto,^{‡,||} and Kian Ping Loh^{*,†,‡}



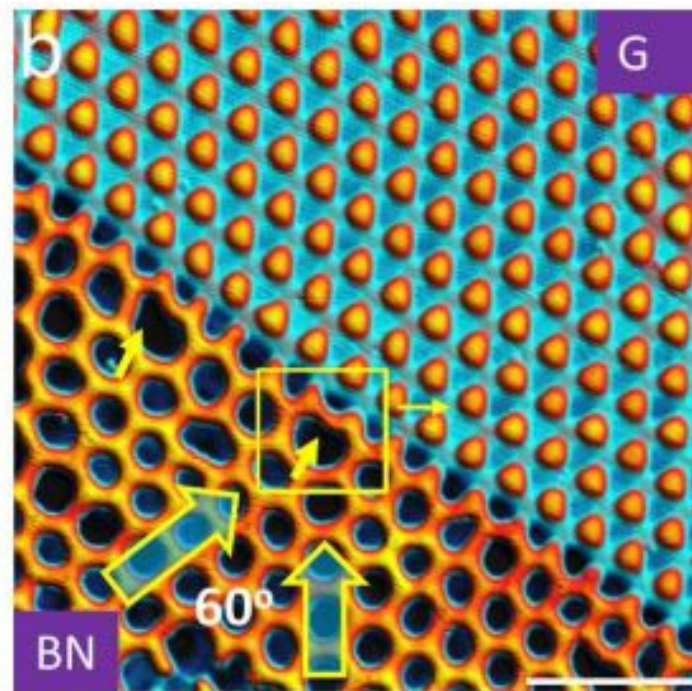
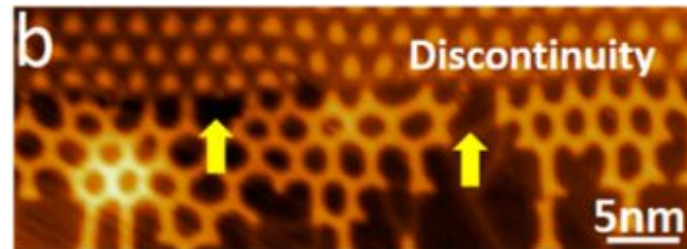
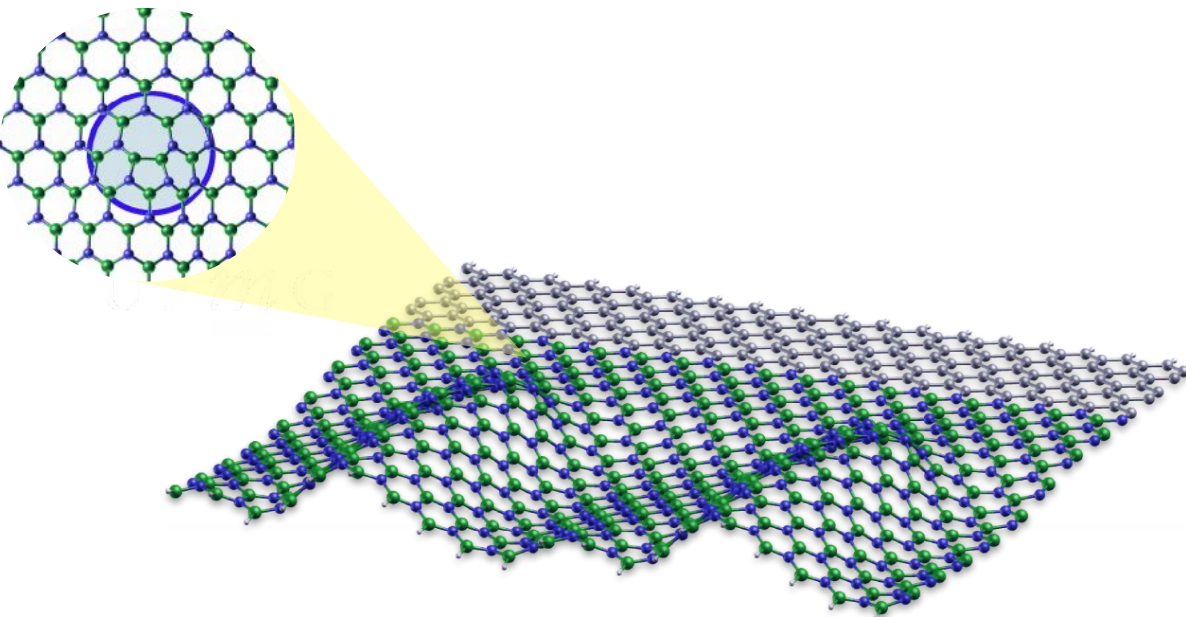
Heteroestructuras

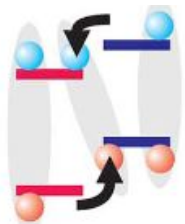
NANO LETTERS

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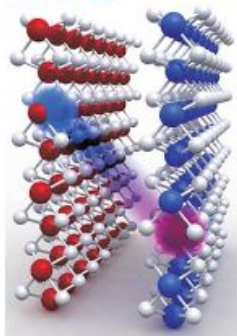
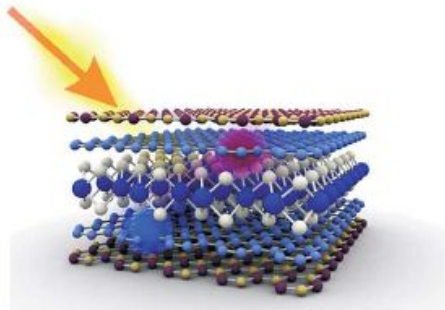
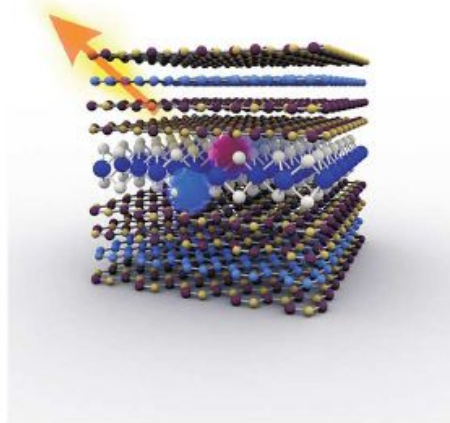
Lattice Relaxation at the Interface of Two-Dimensional Crystals: Graphene and Hexagonal Boron-Nitride

Jiong Lu,^{†,‡,#} Lídia C. Gomes,^{‡,§,#} Ricardo W. Nunes,[§] A. H. Castro Neto,^{‡,||} and Kian Ping Loh^{*†,‡}



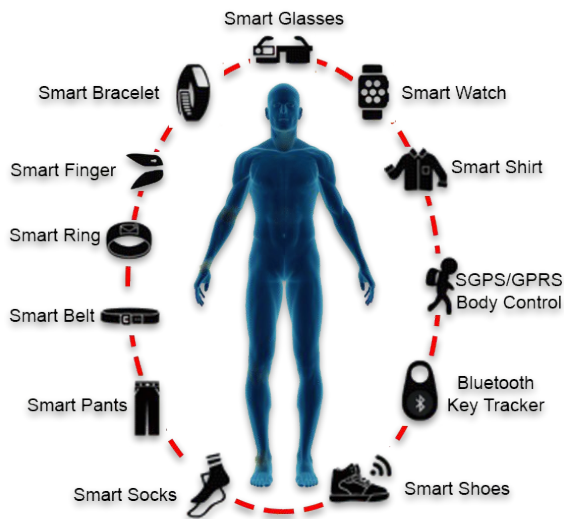
E**F**

MoS_2 WSe_2

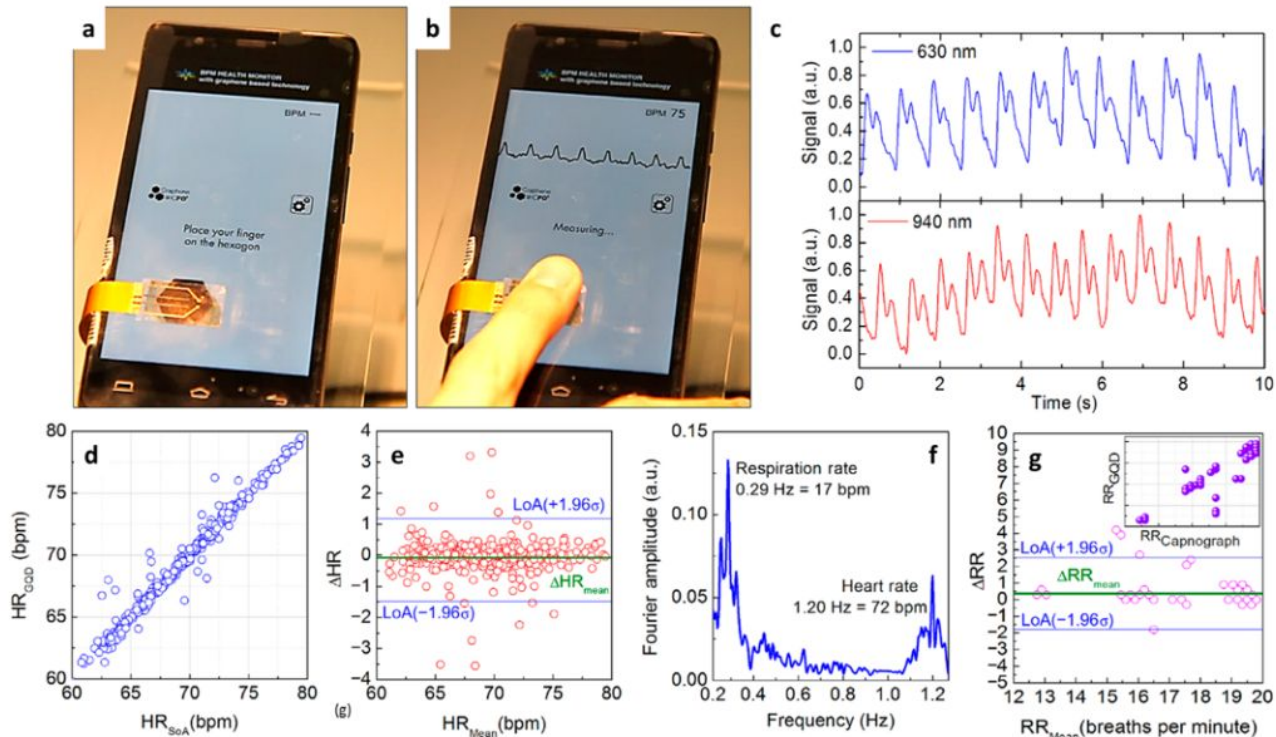
**G****H**

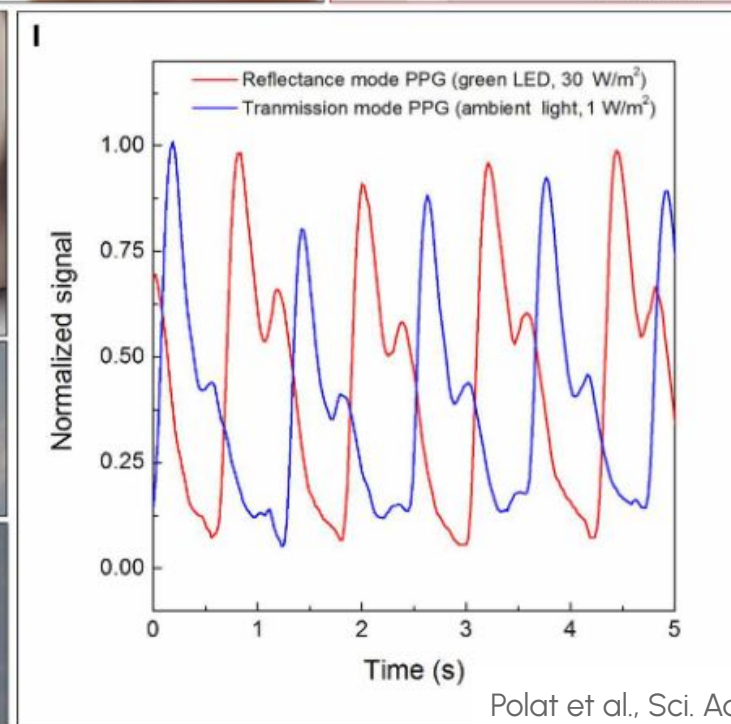
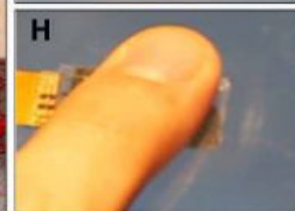
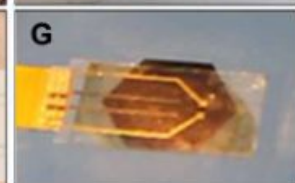
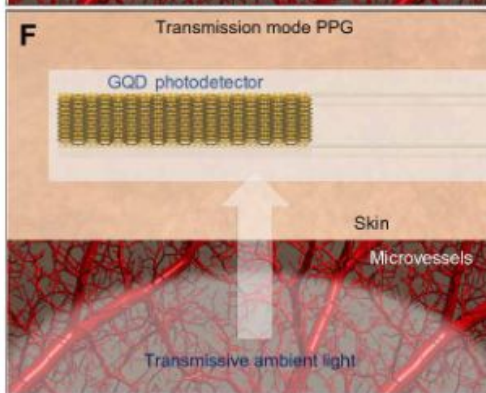
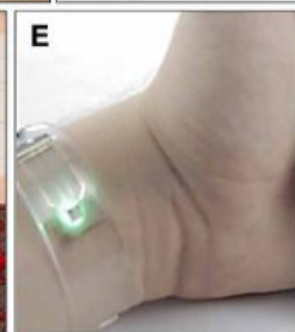
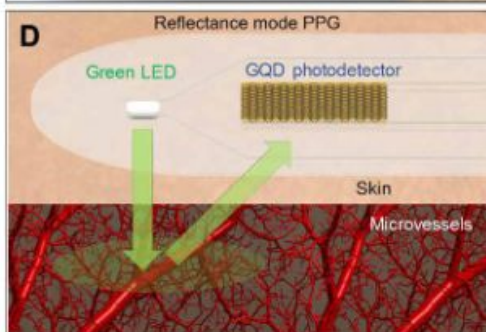
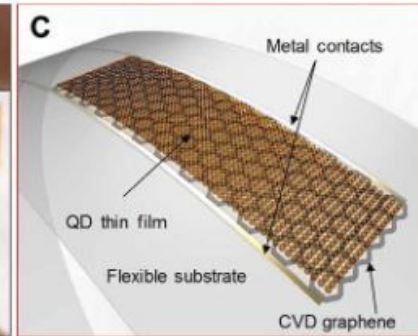
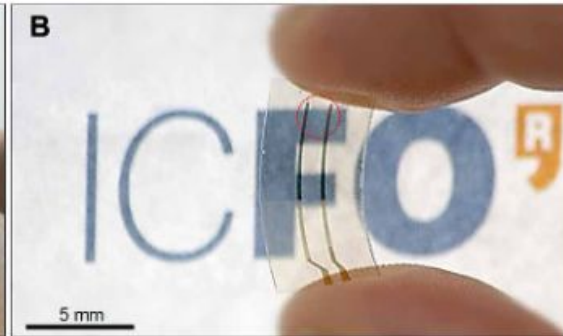
Photodetectors for Biomedical Applications

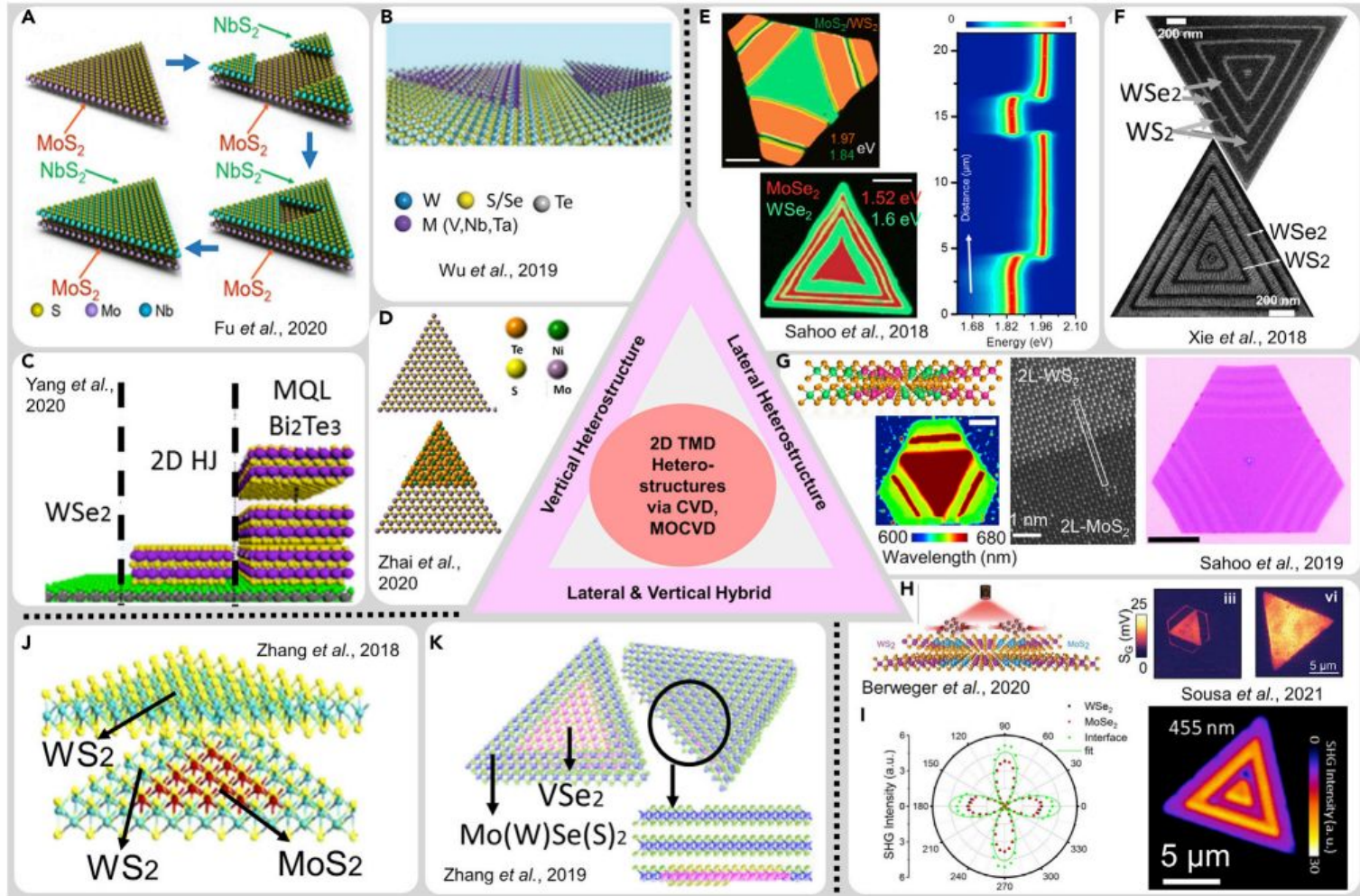
Light-sensitive quantum dots (QDs) made of PbS on the graphene layer.



Wearable technology







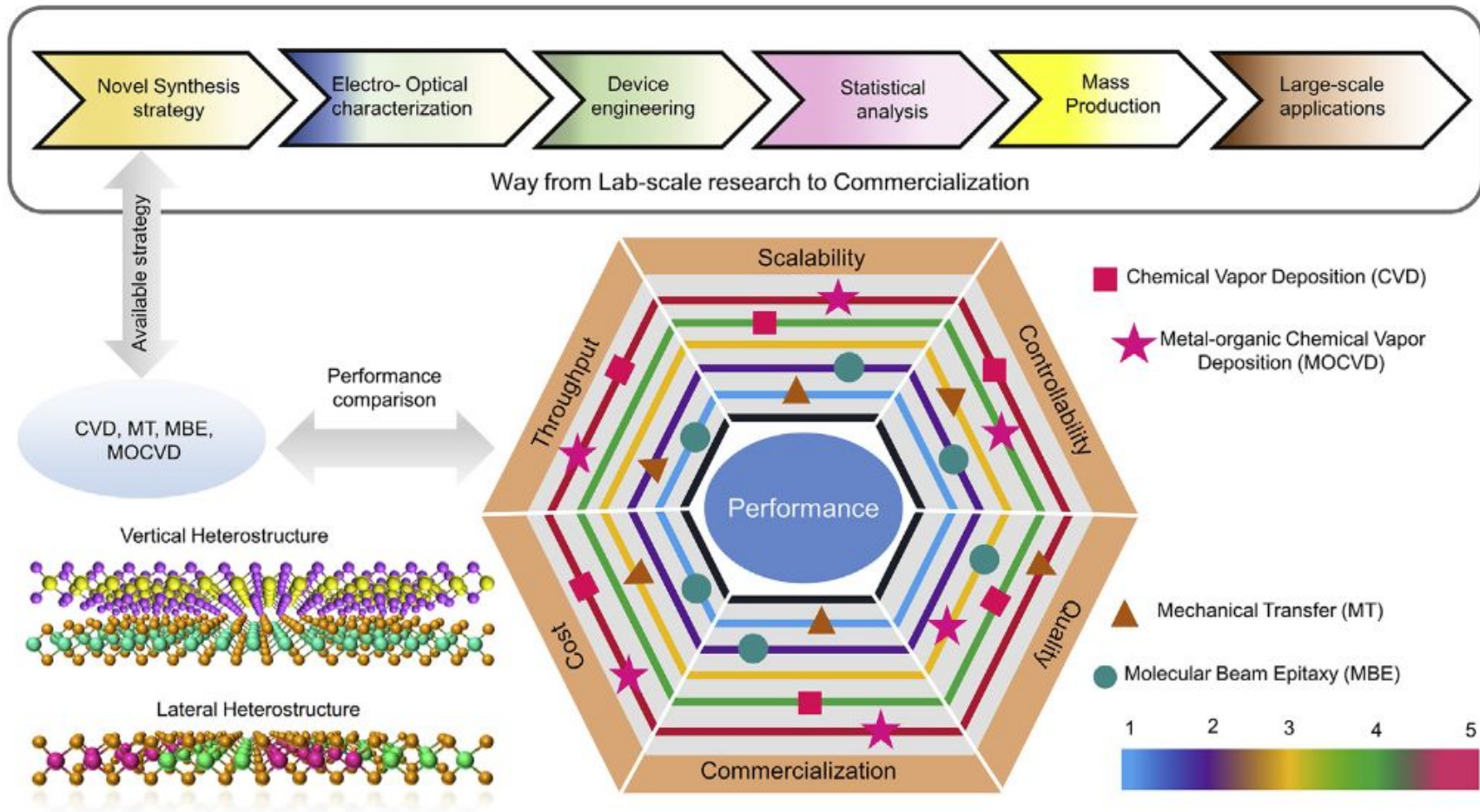
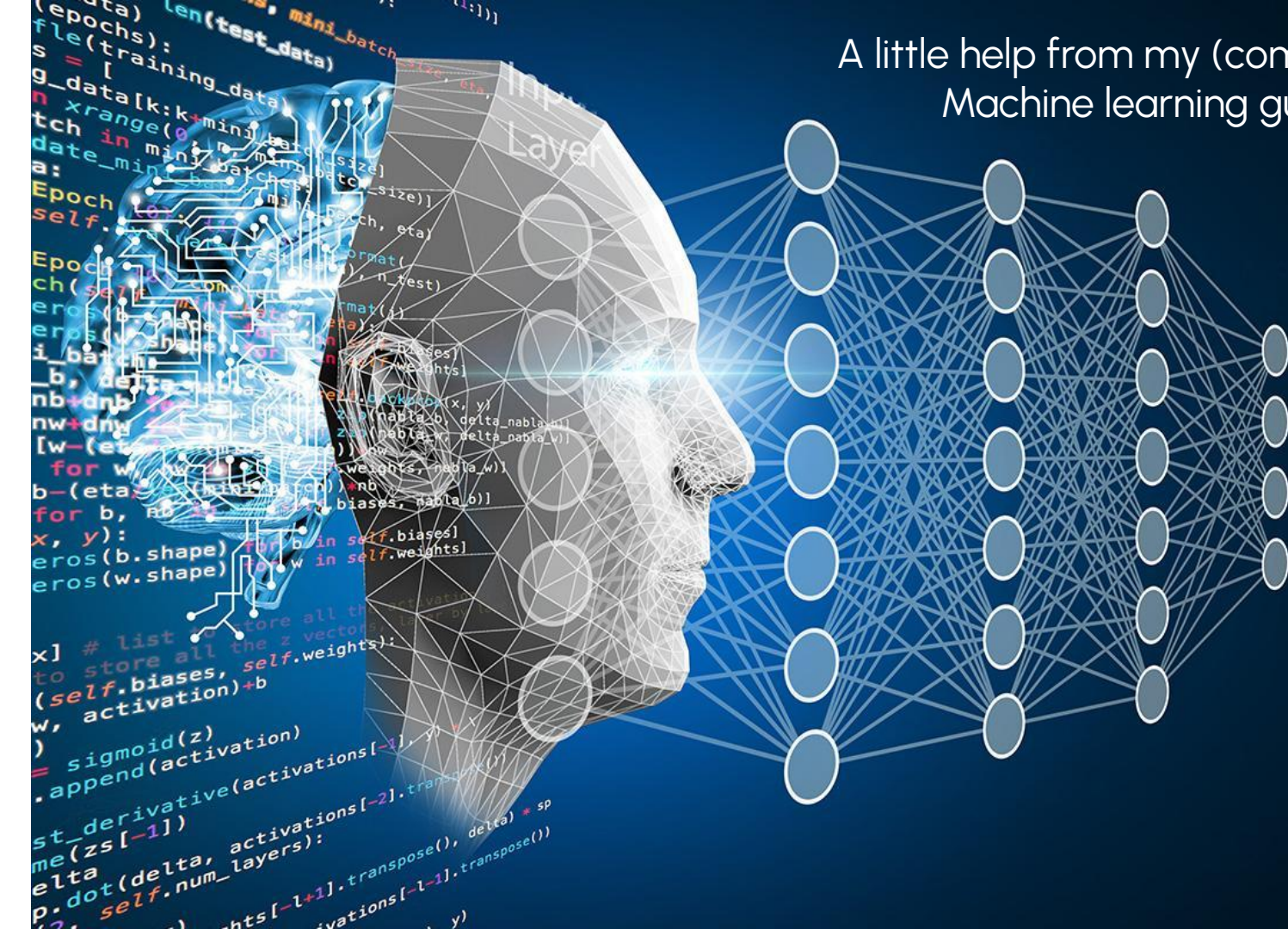


Figure 1. Roadmap toward commercialization

A little help from my (computer) friends:
Machine learning guided synthesis





Machine Learning Driven Synthesis of Few-Layered WTe_2 with Geometrical Control

Manzhang Xu, Bijun Tang, Yuhao Lu, Chao Zhu, Qianbo Lu, Chao Zhu, Lu Zheng, Jingyu Zhang, Nannan Han, Weidong Fang, Yuxi Guo, Jun Di, Pin Song, Yongmin He, Lixing Kang, Zhiyong Zhang*, Wu Zhao, Cuntai Guan, Xuewen Wang*, and Zheng Liu*

Cite this: *J. Am. Chem. Soc.* 2021, 143, 43, 18103–18113

Publication Date: October 4, 2021
<https://doi.org/10.1021/jacs.1c06786>

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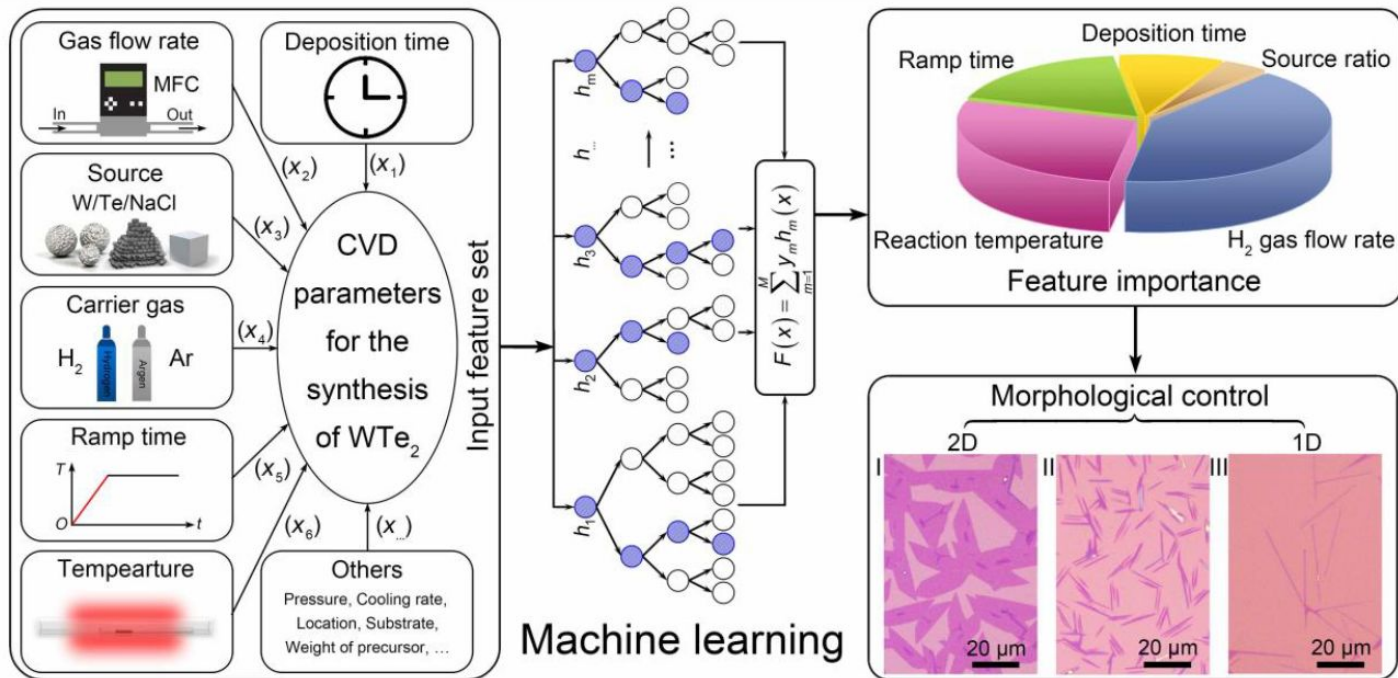
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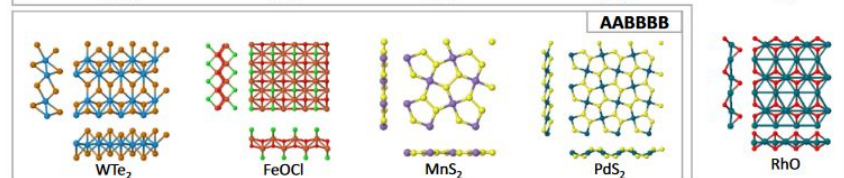
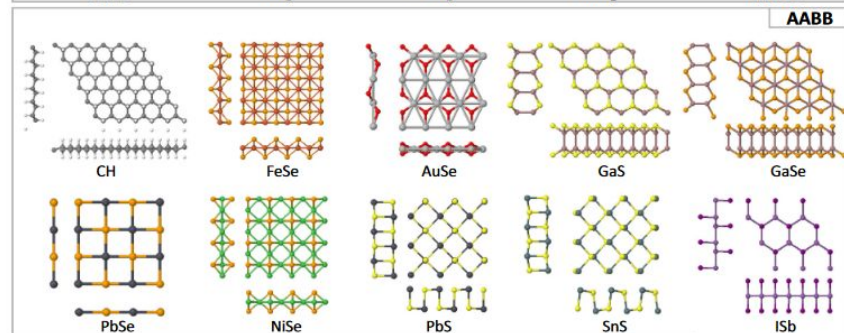
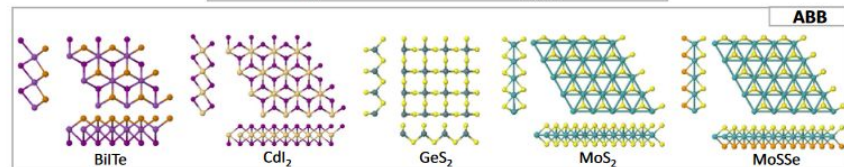
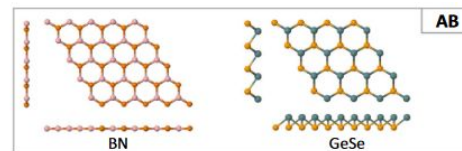
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An artificial intelligence-aided virtual screening recipe for two-dimensional materials discovery

Murat Cihan Sorkun^{1,2,3}, Séverin Astruc^{1,2}, J. M. Vianney A. Koelman^{2,3} and Süleyman Er^{1,2,3*}

In recent years, artificial intelligence (AI) methods have prominently proven their use in solving complex problems. Across science and engineering disciplines, the data-driven approach has become the fourth and newest paradigm. It is the burgeoning of findable, accessible, interoperable, and reusable (FAIR) data generated by the first three paradigms of experiment, theory, and simulation that has enabled the application of AI methods for the scientific discovery and engineering of compounds and materials. Here, we introduce a recipe for a data-driven strategy to speed up the virtual screening of two-dimensional (2D) materials and to accelerate the discovery of new candidates with targeted physical and chemical properties. As a proof-of-concept, we generate new



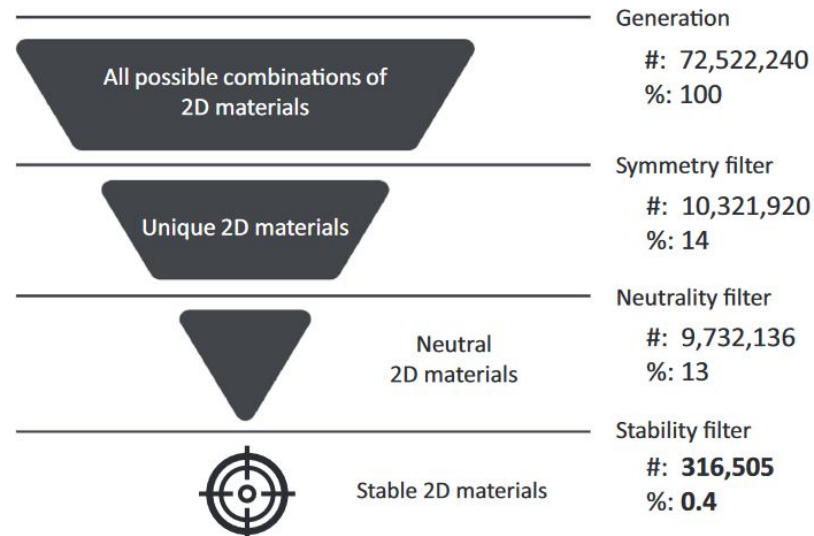
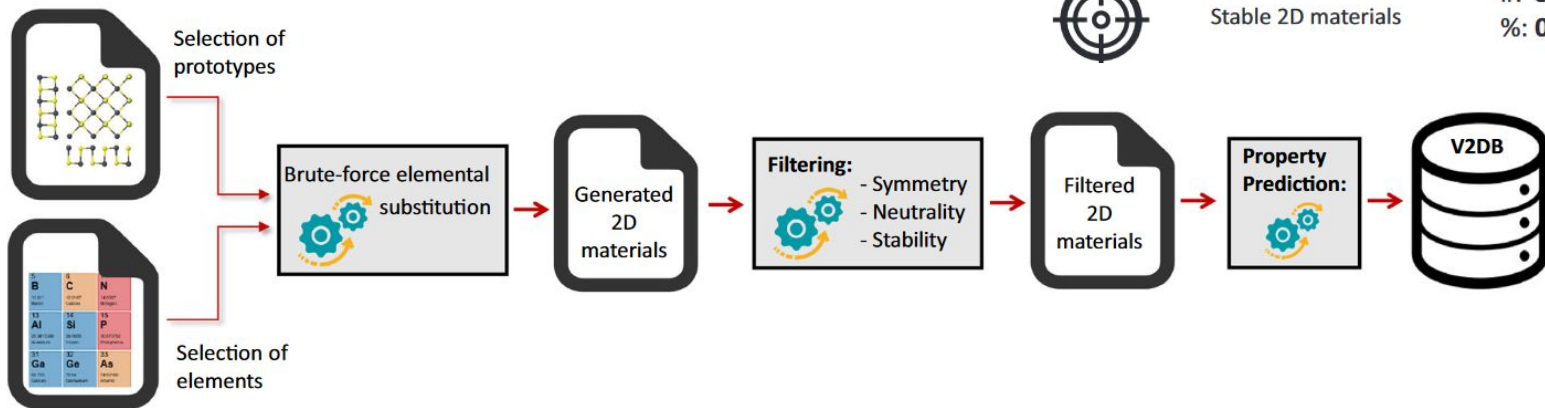
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	H																	He	
	Hydrogen																		Helium
II	3	4																	10
	Li	Be																	Ne
	Lithium	Beryllium																	Neon
III	11	12																	18
	Na	Mg																	Ar
	Sodium	Magnesium																	Argon
IV	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	
	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
	Potassium	Calcium	Scandium	Titanium	Vanadium	Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc	Gallium	Germanium	Arsenic	Selenium	Bromine	Krypton	
V	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	
	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
	Rubidium	Strontium	Yttrium	Zirconium	Niobium	Molybdenum	Technetium	Ruthenium	Rhodium	Palladium	Silver	Cadmium	Indium	Tin	Antimony	Tellurium	Iodine	Xenon	
VI	55	56		72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	
	Cs	Ba	LA	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	
	Cesium	Barium	Lanthanum	Hafnium	Tantalum	Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	Bismuth	Polonium	Astatine	Radon	

Legend: ■ A ■ B ■ Not Used

An artificial intelligence-aided virtual screening recipe for two-dimensional materials discovery

Murat Cihan Sorkun^{1,2,3}, Séverin Astruc^{1,2}, J. M. Vianney A. Koelman^{2,3} and Süleyman Er^{1,2,3*}

In recent years, artificial intelligence (AI) methods have prominently proven their use in solving complex problems. Across science and engineering disciplines, the data-driven approach has become the fourth and newest paradigm. It is the burgeoning of findable, accessible, interoperable, and reusable (FAIR) data generated by the first three paradigms of experiment, theory, and simulation that has enabled the application of AI methods for the scientific discovery and engineering of compounds and materials. Here, we introduce a recipe for a data-driven strategy to speed up the virtual screening of two-dimensional (2D) materials and to accelerate the discovery of new candidates with targeted physical and chemical properties. As a proof of concept, we generate new




Universal machine learning aided synthesis approach of two-dimensional perovskites in a typical laboratory

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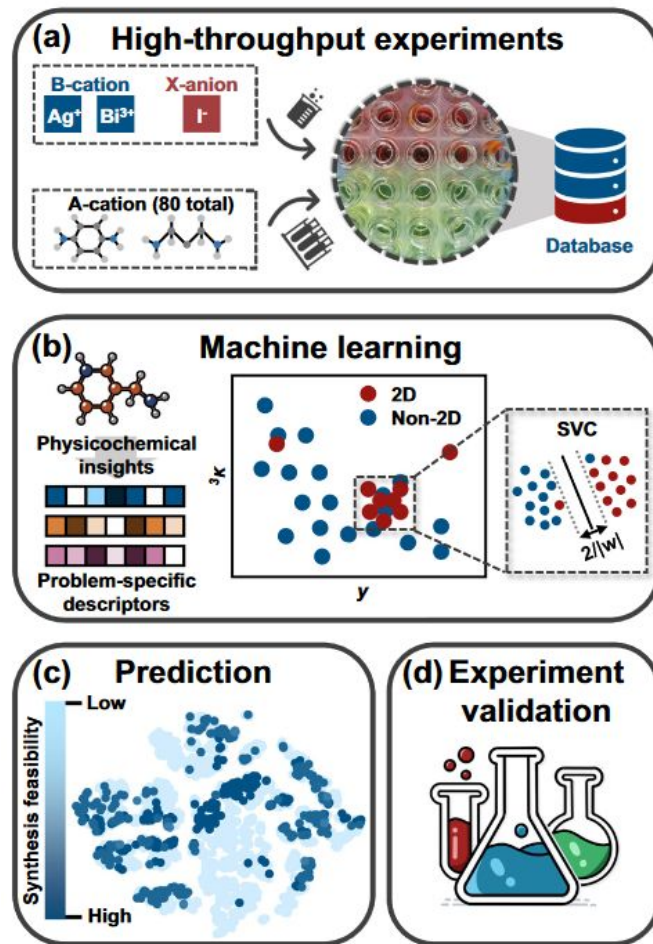
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The past decade has witnessed the significant efforts in novel material discovery in the use of data-driven techniques, in particular, machine learning



Accelerated search for 2D perovskites by machine learning

Predicting Van der Waals Heterostructures by a Combined Machine Learning and Density Functional Theory Approach

Daniel Willhelm, Nathan Wilson, Raymundo Arroyave, Xiaoning Qian, Tahir Cagin*, Ruth Pachter*, and Xiaofeng Qian*

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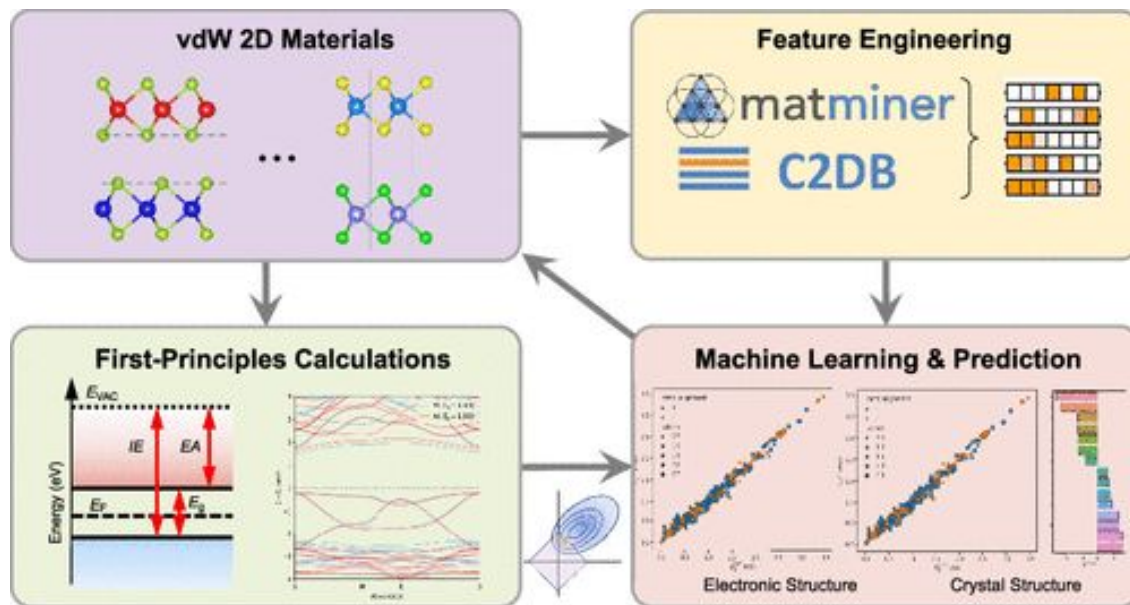
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Deep Learning Enabled Strain Mapping of Single-Atom Defects in Two-Dimensional Transition Metal Dichalcogenides with Sub-Picometer Precision

Chia-Hao Lee, Abid Khan,[#] Di Luo,[#] Tatiane P. Santos, Chuqiao Shi, Blanka E. Janicek, Sangmin Kang, Wenjuan Zhu, Nahil A. Sobh, André Schleife, Bryan K. Clark, and Pinshane Y. Huang*



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ABSTRACT: Two-dimensional (2D) materials offer a platform to study the strain fields induced by individual defects, yet challenges associated with radiation damage limited electron microscopy methods to probe these strain fields. Here, we demonstrate an approach to probe atom defects with sub-picometer precision in a two-dimensional transition metal dichalcogenide, $WSe_{2-2x}Te_{2x}$. We use deep learning to mine large data sets of aberration-corrected transmission electron microscopy images to locate individual point defects. By combining hundreds of images of identical defects, we generate high signal-to-noise class

